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# CLEAN DEVELOPMENT MECHANISM PROJECT DESIGN DOCUMENT FORM (CDM-PDD) Version 03 - in effect as of: 28 July 2006

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### SECTION A. General description of project activity

### A.1 Title of the project activity:

Fosfertil Piaçaguera NAP 2 Nitrous Oxide Abatement Project Version 2

19 September 2007

### **A.2.** Description of the <u>project activity</u>:

Nitrous Oxide ( $N_2O$ ) is an undesired by-product gas from the manufacture of nitric acid. Nitrous oxide is formed during the catalytic oxidation of Ammonia. Over a suitable catalyst, a maximum 98% (typically 92-96%) of the fed Ammonia is converted to Nitric Oxide (NO). The remainder participates in undesirable side reactions that lead to the production of Nitrous Oxide, among other compounds.

Waste  $N_2O$  from nitric acid production is typically released into the atmosphere, as it does not have any economic value or toxicity at typical emission levels.  $N_2O$  is an important greenhouse gas, which has a high Global Warming Potential (GWP) of 310.

The project activity involves the installation of a secondary catalyst to abate N<sub>2</sub>O inside the reactor once it is formed.

The baseline scenario is determined to be the release of  $N_2O$  emissions to the atmosphere at the currently measured rate, in the absence of regulations to restrict  $N_2O$  emissions. If regulations on  $N_2O$  emissions are introduced during the crediting period, the baseline scenario shall be adjusted accordingly.

Baseline emissions rate will be determined by measuring  $N_2O$  emission factor (kg  $N_2O$ /tonnne HNO<sub>3</sub>) during a *complete* production campaign prior to project implementation. To assure that the data obtained during the initial  $N_2O$  measurement campaign for baseline emission factor determination are representative of the actual GHG emissions from the source plant, a set of process parameters known to affect  $N_2O$  generation that are under the control of the plant operator, will be controlled from historical data.

Baseline emissions will be dynamically adjusted from activity levels on an ex-post basis through monitoring the amount of nitric acid production. Project  $N_2O$  emission will be monitored directly in real time. Additional  $N_2O$  monitoring and recording facilities will be installed to measure the amount of  $N_2O$  emitted by the project activity.

Project additionality is determined using the most recent version of the "tool for demonstration and assessment of additionality", approved by the CDM Executive Board.

The project activity will contribute to the sustainable development of the country through industrial technology transfer (catalyst technology from a developed country to Brazil). The project activity will reduce  $N_2O$  emissions and will not increase nor decrease direct emissions of other air pollutants.



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The project does not impact on the local communities or access of services in the area. The project activity will not cause job losses at Fosfertil Piaçaguera NAP (Nitric Acid Plant) 2. Fosfertil Nitrous Oxide Abatement Project has the potential to be replicated by other nitric acid plants in the country and in other developing countries.

### A.3. Project participants:

Name of Party involved (*). ((host) indicates a host Party)	Private and/or public entity(ies) project participants (*) (as applicable)	Kindly indicates if the Party involved wishes to be considered as project participant (Yes/No)
Brazil (host)	Ultrafertil S/A Private entity. Project Developer.	No
Switzerland	Ecoinvest Carbon S.A.	No

<sup>(\*)</sup> In accordance with the CDM modalities and procedures, at the time of making the CDM-PDD public at the stage of validation, a Party involved may or may not have provided its approval. At the time of requesting registration, the approval by the Party(ies) involved is required.

With an operational infrastructure that includes its own mines, processing plants and industrial processing units, in addition to an efficient logistics system, Fosfertil is the Brazil's main producer of phosphated and nitrogened raw materials for fertilizer production.

Currently, Fosfertil has approximately 2,600 employees in four different Brazilian states. The chemical plants are located in Araucária (State of Paraná), Cubatão (State of São Paulo), Piaçagüera, in the city of Cubatão (São Paulo) and Uberaba (State of Minas Gerais).

The mining facilities are listed in Catalão (State of Goiás), Tapira and Patos de Minas (Minas Gerais) – Catalão and Patos de Minas also have industrial units for production of phosphated inputs. Fosfertil also counts with a Marine Terminal in Santos (São Paulo), a geology research unit in Patrocínio (MG) and the São Paulo Headquarters.

In all these places Fosfértil operates in an ethical, responsible way, seeking to make partnerships with the communities aimed at obtaining social and economic benefits.

### **Our History**

1958: The Fertilizer Factory of Cubatão (Fafer) is inaugurated.

1965: With participation of Philips/PS Petroleum, Grupo Ultra and international financial entities, Ultrafertil S/A is constituted in Cubatão (São Paulo).

1969: Ultrafertil's Marine Terminal starts its operations in Santos (São Paulo).





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1970: Ultrafertil inaugurates its fertilizer industrial facility in Piaçagüera, Cubatão.

1974: Petrobras purchases Ultrafertil's shareholding control.

1977: Fertilizantes Fosfatados S/A – Fosfertil is created as a State-owned company to exploit the Patos de Minas (MG) phosphate rock deposit.

1977: Fafer is merged into Ultrafertil.

1980: Fosfertil incorporates Valep, a phosphate mining company in Tapira (Minas Gerais), and Valefertil, a fertilizer chemical complex in Uberaba (Minas Gerais).

1982: A new Ultrafertil's industrial plant starts operating in Araucária (Paraná).

1992: Under the National Program of Destatization, Fosfertil is privatized and its shareholding control is acquired by the Fertifos Consortium, a group of fertilizer-industry companies.

1992: Fosfertil becomes a publicly-traded company, with shares traded on the stock exchange.

1993: Ultrafertil's privatization auction is carried out. The company is acquired by Fosfertil.

1995: Ultrafertil is merged into Goiasfertil, a wholly-owned subsidiary of Fosfertil dedicated to the extraction of phosfhate rock in Catalão (Goiás). After the merger, Goiasfertil's corporate name changes to Ultrafertil S/A.

2004: Even maintaining the corporate names Fertilizantes Fosfatados S/A - Fosfertil and Ultrafertil S/A, the company decides to adopt a single corporate identity — Fosfertil — and updates its logo. Therefore Ultrafertil S/A is the Corporate Name and Fosfertil is the Trade Mark Name.





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A.4. Tecl	.4. Technical description of the <u>project activity</u> :		
A.4.1. Location of the <u>project activity</u> :			
	A.4.1.1.	Host Party(ies):	
Brazil			
	A.4.1.2.	Region/State/Province etc.:	
Sao Paulo			
	A.4.1.3.	City/Town/Community etc:	

Cubatão/Jardim São Marcos

A.4.1.4. Detail of physical location, including information allowing the unique identification of this <u>project activity</u> (maximum one page):

The project activity is located at Fosfertil Piaçaguera NAP 2 located in Estrada Eng. Plínio de Queiroz S/N, Cubatão (23°56' South; 46°20' west), state of Sao Paulo, Brazil.

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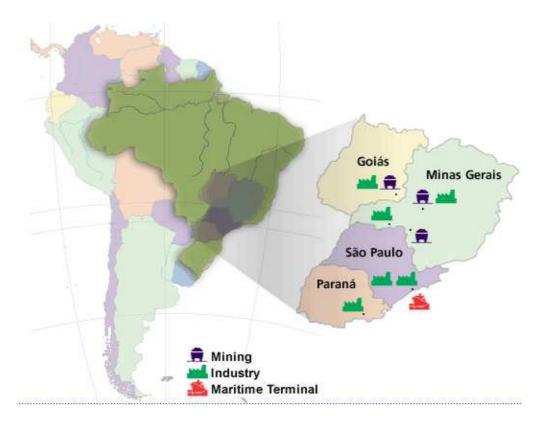


Figure 1. Location of Fosfertil's facilities in Brazil.

### A.4.2. Category(ies) of project activity:

The project activity fall within Sectoral scope: "(5) Chemical industries".

### A.4.3. Technology to be employed by the project activity:

### The Ostwald process

Nowadays, all commercial Nitric Acid is produced by the oxidation of ammonia, and subsequent reaction of the oxidation products with water, through the Ostwald process.

The basic Ostwald process involves 3 chemical steps:

A) Catalytic oxidation of ammonia with atmospheric oxygen, to yield Nitrogen Monoxide (or Nitric Oxide).

(1) 
$$4 \text{ NH}_3 + 5 \text{ O}_2 \rightarrow 4 \text{ NO} + 6 \text{ H}_2\text{O}$$



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- B) Oxidation of the Nitrogen Monoxide to Nitrogen Dioxide or Dinitrogen Tetroxide
  - (2)  $2 \text{ NO} + \text{O}_2 \rightarrow 2 \text{ NO}_2 \rightarrow \text{N}_2\text{O}_4$
- C) Absorption of the Nitrogen Oxides with water to yield Nitric Acid
  - (3)  $3 \text{ NO}_2 + \text{H}_2\text{O} \rightarrow 2 \text{ HNO}_3 + \text{NO}$

Reaction 1 is favored by lower pressure and higher temperature. Nevertheless, at too high temperature, secondary reactions take place that lower yield (affecting nitric production); then, an optimal is found between 850-950 C, affected by other process conditions and catalyst chemical composition (figure 2)<sup>1</sup>. Reactions 2 and 3 are favored by higher pressure and lower temperatures.

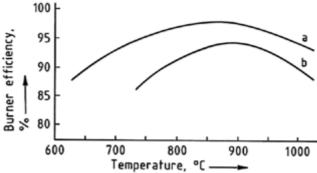


Figure 2. Conversion of Ammonia to Nitrogen Monoxide on Platinum Gauze as a function of temperature a) 100 kPa; (b) 400 kPa [1]

The way in which these three steps are implemented, characterizes the various Nitric Acid processes found throughout the industry. In mono pressure or single pressure processes ammonia combustion and nitrogen oxide absorption take place at the same working pressure. In dual pressure or split pressure plants the absorption pressure is higher than the combustion pressure.

### **Nitrous Oxide formation**

Nitrous oxide is formed during the catalytic oxidation of Ammonia. Over a suitable catalyst, a maximum 98% (typically 92-96%) of the fed Ammonia is converted to Nitric Oxide (NO) according to reaction (1) above. The remainder participates in undesirable side reactions that lead to Nitrous Oxide ( $N_2O$ ), among other compounds.

Side reactions during oxidation of Ammonia:

(4)  $4 \text{ NH}_3 + 4 \text{ O}_2 \rightarrow 2 \text{ N}_2\text{O} + 6 \text{ H}_2\text{O}$  (Nitrous Oxide formation).

(5) 
$$4 \text{ NH}_3 + 3 \text{ O}_2 \rightarrow 2 \text{ N}_2 + 6 \text{ H}_2\text{O}$$

<sup>&</sup>lt;sup>1</sup> Thieman et al., "Nitric Acid, Nitrous Acid, and Nitrogen Oxides", *Ullmann's Encyclopedia of Industrial Chemistry 6th Edition*, Wiley-VCH Verlag GmbH & Co. KGaA. All rights reserved.



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- (6)  $2 \text{ NO } \rightarrow \text{N}_2 + \text{O}_2$
- (7)  $4 \text{ NH}_3 + 6 \text{ NO} \rightarrow 5 \text{ N}_2 + 6 \text{ H}_2\text{O}$

### N<sub>2</sub>O abatement technology classification

The potential technologies (proven and under development) to treat  $N_2O$  emissions at Nitric acid plants, have been classified as follows, based on the process location of the control device:

Primary: N<sub>2</sub>O is prevented from forming in the oxidation gauzes.

Secondary: N<sub>2</sub>O once formed, is eliminated anywhere between the outlet of the ammonia oxidation gauzes and the inlet of the absorption tower.

Tertiary:  $N_2O$  is removed at the tail gas, after the absorption tower and previous to the expansion turbine. Quaternary:  $N_2O$  is removed following the expansion turbine, and before the stack.

### Selected technology for the project activity

### General description

The current project activity involves the installation of a new (not previously installed) catalyst below the oxidation gauzes (a "secondary catalyst") whose sole purpose is the decomposition of  $N_2O$ ; the secondary approach has the following advantages:

- The catalyst does not consume electricity, steam, fuels or reducing agents (all sources of leakage) to eliminate  $N_2O$  emissions; thus, operating costs are negligible and the overall energy balance of the plant is not affected.
- Installation is extremely simple and does not require any new process unit or re-design of existing ones (only in a few cases, the reactor basket needs some minor modifications to accommodate the new catalyst).
- Installation is also very fast, so it is done simultaneously with a primary gauze changeover; thus, the plant has no loss in production due to incremental down time.
- Considerably lower capital cost when compared to other approaches.

Fosfertil is determined to install a secondary catalyst system (upon successful registration as a CDM project) and have selected Johnson Matthey Technology.

Johnson Matthey has been developing solutions for a "secondary" catalyst whose sole purpose is to decompose  $N_2O$  without affecting Nitric Acid production. Typically the catalyst has a very high activity for  $N_2O$  decomposition (minimum of 80% of abatement can be reached). Beyond high abatement of  $N_2O$ , some other advantages of the use of secondary catalyst are: proven performance, no measurable effect on ammonia to nitric oxide yield, and its implementation does not lead to increased  $NO_X$  emissions.

The catalyst is placed directly after the platinum catalyst, partially replacing the supports (Raschig-rings) of that catalyst. In this case, retrofit is relatively simple. Fosfertil Piaçaguera NAP 2 has a Raschig-rings bed as part of the support/homogenization of system on their oxidation reactors. Due to its high degree of selectivity (towards N<sub>2</sub>O decomposition) the depth of the secondary catalyst bed to be installed is only a





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few centimetres thick, thinner than the actual Raschig-rings bed. To create space to insert the new catalyst, enough layers of Raschig rings will be removed from the basket. Once the secondary catalyst is installed, the primary gauzes are placed on top of the basket, as usual. Then, the secondary catalyst acts as support system for the primary gauze pack and both catalysts are in close contact.

The chosen  $N_2O$  abatement catalyst vendor will be obligated by Fosfertil to guarantee a minimum of 80% of  $N_2O$  decomposition, as well as to take back the catalyst at the end of their useful life and refine, recycle or dispose of it according to the prevailing standards and hence fulfil sustainability standards

Once installed, the catalyst itself and the AMS will be operated by the local Fosfertil Piaçaguera NAP 2 employees. All project participants will work together on training Fosfertil Piaçaguera NAP 2 workers to reliably supervise the effective operation of the catalyst technology, apply the installed monitoring system to measure the emissions levels and collect the data in a manner that allows the successful completion of each verification procedure.

### A.4.4 Estimated amount of emission reductions over the chosen crediting period:

Total *ex-ante* emissions reductions are estimated to be 171,931 tonnes CO<sub>2</sub>e/year for the first seven-year crediting period, which may be renewed. Note that actual emissions reductions will be based on monitored data and may differ from this estimate.

Year	Annual estimation of emission reductions in tonnes of CO <sub>2</sub> e
	Plant 2
	Piaçaguera
$2008^2$	150,440
2009	171,931
2010	171,931
2011	171,931
2012	171,931
2013	171,931
2014	171,931
2015 <sup>3</sup>	21,491
<b>Total estimated reductions</b> (tonnes of CO <sub>2</sub> e)	1,203,517
Total number of crediting years	7
Annual average over the crediting period of estimated reductions	171,931
(tonnes of CO <sub>2</sub> e)	

### A.4.5. Public funding of the <u>project activity</u>:

No funds from public national or international sources are involved in any aspect of the proposed project.

<sup>&</sup>lt;sup>2</sup> Note: Year 2008 includes ten months and a half, from Middle of February to December

<sup>&</sup>lt;sup>3</sup> Year 2015 includes one month and a half, from January to middle of February



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### SECTION B. Application of a baseline and monitoring methodology

# B.1. Title and reference of the <u>approved baseline and monitoring methodology</u> applied to the <u>project activity</u>:

The selected methodology is AM0034 "Catalytic reduction of N<sub>2</sub>O inside the ammonia burner of nitric acid plants" version 02. (EB 27)

AM0028 "Catalytic reduction of N<sub>2</sub>O in the tail gas of Nitric Acid or Caprolactam Productions Plants" version 04.1 (EB 28) is used to select the baseline scenario".

The "Tool for the demonstration and assessment of additionality" version 03 (EB 29) is used to demonstrate additionality.

# **B.2** Justification of the choice of the methodology and why it is applicable to the <u>project</u> activity:

The proposed project activity would reduce N<sub>2</sub>O emissions from Fosfertil NAP 2 in Piaçaguera Site meeting all the conditions specified in the selected approved methodology (AM0034):

- Fosfertil Piaçaguera NAP 2 limits the application of this project activity to existing nitric acid production capacity of 95,800 ton HNO<sub>3</sub>/year installed in 1989. The Piaçaguera NAP 2 was installed in 1989 with 220 ton/d of design capacity but this nitric acid production capacity was enlarged before 31 December 2005, obtaining the current number.
- Currently Fosfertil Piaçaguera NAP 2 doesn't have any N<sub>2</sub>O destruction or abatement facility or equipment that could be affected by the project activity;
- The project activity will not affect the level of nitric acid production
- There are currently no regulatory requirements or incentives to reduce levels of N<sub>2</sub>O emissions from nitric acid plants in Brazil.
- As it was explained above, no N<sub>2</sub>O abatement technology is currently installed in Fosfertil NAP 2 at Piaçaguera Site.
- The secondary catalyst technology to be installed as project activity has been tested in several industrial trials and has been demonstrated that its installation does not increase NO<sub>x</sub> emissions.
- The NO<sub>X</sub> abatement catalyst system installed in the unit, prior the start of the project activity, is a SCR, Selective Catalytic Reduction
- As it was explained before, the secondary catalyst technology to be installed as project activity
  has been tested in several industrial trials and has been demonstrated that its operation does not
  lead to any process emissions of greenhouse gases, directly or indirectly.
- Continuous real-time measurements of N<sub>2</sub>O concentration and total gas volume flow will be carried out in the stack:
  - o Prior to the installation of the secondary catalyst for one campaign, and
  - o After the installation of the secondary catalyst throughout the chosen crediting period of the project activity.

# B.3. Description of the sources and gases included in the project boundary





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The project boundary encompasses the physical, geographical site of Fosfertil NAP 2 at Piaçaguera Site and equipment for the complete nitric acid production process from the inlet to the ammonia burner to the stack. The only GHG emission relevant to the project activity is  $N_2O$  contained in the waste stream exiting the stack. The abatement of  $N_2O$  is the only GHG emission under the control of the project participant.

The secondary catalyst utilizes the heat liberated by the highly exothermal oxidation reaction (that occurs on the precious metal gauzes of the primary catalyst) to reach its effective operating temperature. Once the operating temperature is reached, no incremental energy is necessary to sustain the reaction.

	Source	Gas	Included?	Justification / Explanation
le		$CO_2$	Excluded	The project does not lead to any change
Baseline	Nitric Acid Plant (Burner Inlet to Stack)		Excluded	in CO <sub>2</sub> or CH <sub>4</sub> emissions, and, therefore, these are not included.
В			Included	
_	Nitric Acid Plant (Burner	$CO_2$	Excluded	The project does not lead to any change
rit,	Inlet to Stack)	$CH_4$	Excluded	in CO <sub>2</sub> or CH <sub>4</sub> emissions
ctiv	mict to Stack)	$N_2O$	Included	
Project Activity	Leakage emissions from	$CO_2$	Excluded	No leakage emissions are expected.
oje	production, transport,	$CH_4$	Excluded	
operation and decommissioning of the catalyst.		N <sub>2</sub> O	Excluded	

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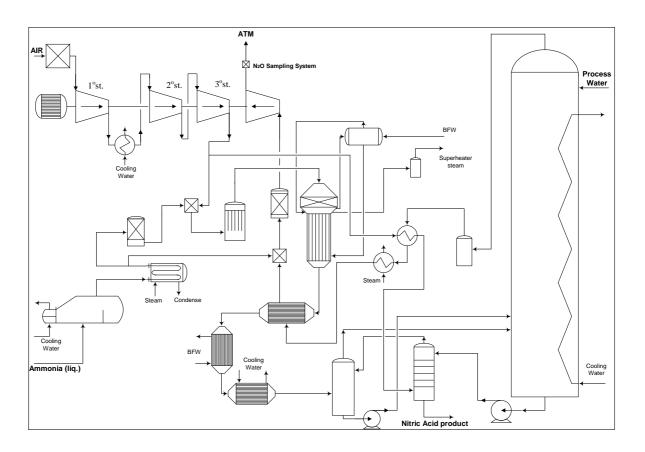


Figure 3. Project boundary of Piaçaguera NAP 2.

# **B.4**. Description of how the <u>baseline scenario</u> is identified and description of the identified baseline scenario:

The baseline methodology application first involves an identification of possible baseline scenarios, and eliminating those that would not qualify. The procedures followed for baseline scenario selection correspond to AM0028 "Catalytic  $N_2O$  destruction in the tail gas of Nitric Acid and Caprolactam Production Plants" Version 04.1 as it is specified in selected AM0034 Version 02. The analysis of baseline scenarios involves six steps:

### Step 1. Identify technically feasible baseline scenario alternatives to the project activity.

The first step in determining the baseline scenario is to analyse all options available to project participants. These include the business-as-usual case, considering sectoral policies and circumstances to



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determine whether this case corresponds to the continuation or not of the current operation of the nitric acid industry, the project scenario, and any other scenarios that might be applicable. This *first step* can be further broken down into two sub-steps:

**Step 1a:** The baseline scenario alternatives should include all possible options that are technically feasible to handle  $N_2O$  emissions. These options include:

- Continuation of *status quo*. The continuation of the current situation, where there will be no installation of technology for the destruction or abatement of  $N_2O$ .
- Switch to alternative production method not involving ammonia oxidation process
- Alternative use of N<sub>2</sub>O, such as:
  - o Recycling N<sub>2</sub>O as a feedstock
  - o Use of N<sub>2</sub>O for external purposes
- The installation of an N<sub>2</sub>O destruction or abatement technology:
  - o Primary approach
  - o Secondary approach
  - o Tertiary approach, including Non Selective Catalytic Reduction (or NSCR De NO<sub>X</sub>)<sup>4</sup>
  - o Quaternary (or end of pipe) approach.

The options include the CDM project activity not implemented as a CDM project.

**Step 1b:** In addition to the baseline scenario alternatives of Step 1a, all possible options that are technically feasible to handle  $NO_X$  emissions should be considered, since some  $NO_X$  technical solutions could also have an effect on  $N_2O$  emissions. The alternatives include:

- The continuation of the current situation, whether a DeNO<sub>X</sub> unit is installed or not
- Installation of a new Extended Absorption tower
- Installation of a new Selective Catalytic Reduction (SCR) DeNO<sub>X</sub> unit
- Installation of a new Non Selective Catalytic Reduction (NSCR) De NO<sub>X</sub> unit
- Installation of a combined NO<sub>X</sub> /N<sub>2</sub>O abatement unit (e.g. UHDEs Envinox® process)
- Installation of a new end-of-pipe treatment such as chemical (H<sub>2</sub>O<sub>2</sub>) scrubbing system

### Step 2: Eliminate baseline alternatives that do not comply with legal or regulatory requirements.

Currently, there are no national regulations or legal obligations in Brazil concerning  $N_2O$  emissions. It is unlikely that any such limits on  $N_2O$  emissions will be imposed in the near future. In fact, given the cost and complexity of suitable  $N_2O$  destruction and abatement technologies, it is unlikely that a limit would be introduced at Brazil considering it has ratified the Kyoto Protocol and actively participates in CDM.

The Fosfertil NAP 2 at Piaçaguera Site has installed a Selective Catalytic Reduction (SCR) DeNOx unit since 1989 and fulfils the Operating License issued by the Environmental Agency of Sao Paulo State (Cetesb). NOx concentration in the stack gas of the nitric acid plant is lower than or equal at 200 ppm volume expressed as NO<sub>2</sub>. Therefore the continuation of the status quo is a valid baseline alternative.

 $<sup>^4</sup>$  NSCR: A NSCR DeNO<sub>X</sub>-unit will reduce  $N_2O$  emissions as a side reaction to the NO<sub>X</sub>--reduction, consequently, new NSCR installation can be seen as an alternative  $N_2O$  reduction technology.



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None of the baseline alternatives can be eliminated in this step because they are all in compliance with legal and regulatory requirements.

### Step 3: Eliminate baseline alternatives that face prohibitive barriers (barrier analysis):

On the basis of the alternatives that are technically feasible and in compliance with all legal and regulatory requirements, a complete list of barriers that would prevent alternatives to occur in the absence of CDM is established.

The identified barriers are:

- Investment barriers, inter alia:
  - Debt funding is not available for this type of innovative project activity;
  - No access to international capital markets due to real or perceived risks associated with domestic or foreign direct investment in the country where the project activity is to be implemented.
- Technological barriers, inter alia:
  - Technical and operational risks of alternatives;
  - Technical efficiency of alternatives (e.g. N<sub>2</sub>O destruction, abatement rate);
  - Skilled and / or properly trained labour to operate and maintain the technology is not available and no education / training institution in the host country provides the needed skill, leading to equipment disrepair and malfunctioning;
  - Lack of infrastructure for implementation of the technology;
- Barriers due to prevailing practice, inter alia:
  - The project activity is the "first of its kind": No project activity of this type is currently operational in the host country or region. (Rhodia Poliamida e Especialidades Ltda, at Paulínia, state of Sao Paulo, has installed a secondary catalyst to eliminate N<sub>2</sub>O emissions, also as CDM project activity).

There are four different groups of  $N_2O$  destruction or abatement technologies at nitric acid plants: primary, secondary, tertiary and quaternary (or end of pipe) measures.

Currently, there is no technology from the primary approach group that reaches high enough removal efficiency, as to represent a potential  $N_2O$  abatement solution in itself.

Available tertiary approaches are the NSCR (Non Selective Catalytic Reduction) and the EnviNO $_X$ ® process commercialized by Uhde GmbH (Germany); both systems are not selective towards N $_2$ O abatement, and also actuate over acidic species (NO $_X$ ). Although Uhde's process is more efficient than the traditional NSCR system, both technologies have significant requirements regarding space and downtime for installation, and consume reducing agents (fuels and/or Ammonia) to attain N $_2$ O abatement (high operating costs). Furthermore, the plant involved in the project has already a DeNO $_X$  system of the SCR type, and then the installation of either technology is partly redundant (the already existing SCR



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would have to be removed and DeNO<sub>X</sub> catalyst disposed of properly). Regardless of these drawbacks, such constraints could be solved from the technical standpoint, so both technologies are considered viable alternatives to abate N<sub>2</sub>O emissions.

The potential removal of  $N_2O$  after the expansion turbine (the quaternary or end-of-pipe approach) has been only studied from the theoretical standpoint and at a laboratory scale. Worldwide, no full scale installations that use such technology are known.

Switch to alternative production method not involving ammonia oxidation process is not an option because there is no other commercially viable alternative to produce nitric acid.

The use of  $N_2O$  for external purposes is technically not feasible at Piaçaguera NAP 2, as the quantity of gas to be treated is extremely high compared to the amount of nitrous oxide that could be recovered. Note,  $N_2O$  concentration in the tail gas at Piaçaguera NAP 2 is expected to be in the range of 500 ppmv to 1,500 ppmv. The use of  $N_2O$  for external purposes is practiced neither in Brazil nor anywhere else.

We may discard recycling  $N_2O$  as a feedstock for the nitric acid plant. This is because nitrous oxide is not a feedstock for nitric acid production. Nitrous oxide is not recycled at nitric acid plants in Brazil, or anywhere else.

Therefore the following baseline alternatives are eliminated in this step:

- Installation of a primary or quaternary N<sub>2</sub>O abatement technology
- The use of  $N_2O$  for external purposes
- Recycling of N<sub>2</sub>O as a feedstock for the plant

Other possible alternatives face no major technological barriers, but require additional investments. These alternatives are considered in Step 4 below.

### Step 4: Identify the most economically attractive baseline scenario alternative:

To conduct the investment analysis, the following sub-steps are used:

**Sub-step 4a**: Determine appropriate analysis method:

Since the project alternatives generate no financial or economic benefits other than CDM related income, then the simple cost analysis should be applied.

### **Sub-step 4b:** Apply simple cost analysis:

The possible alternatives listed in Step 1a above, and not discarded in the barrier analysis stage, involve the installation of some form of secondary or tertiary  $N_2O$  destruction or abatement technology. Both approaches involve substantial investment, and would need to provide benefits other than CDM revenue in order to qualify as valid baselines. Furthermore, tertiary technologies have incremental environmental costs for Fosfertil, since both consume fuels and/or reducing agents to operate, and their installation require the removal of the existing  $DeNO_X$  system which catalyst would have to be disposed of properly.



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No income from any kind of potential product or by-product except CERs are able to pay back investment costs as well as running costs for the installation of any available secondary or tertiary abatement systems as no marketable products or by-products are generated by these  $N_2O$  treatment methods.

According to the baseline methodology,

"If all alternatives do not generate any financial or economic benefits, then the least costly alternative among these alternative pre-selected as the most plausible baseline scenario."

As a result the only feasible baseline is a continuation of the *status quo*, which meets current regulations, and requires neither additional investments nor additional running costs.

Therefore the continuation of the current situation can be pre-selected as the baseline scenario.

**Sub-step 4c** is not applied, since a simple cost analysis is adequate for this project.

Sub-step 4d: Sensitivity analysis

Since the economic analysis is based on simple cost analysis, the baseline methodology does not require a sensitivity analysis: the results are not sensitive to such factors as inflation rate, investment costs, etc. since there are no economic benefits.

**Step 5**: Re-assessment of Baseline Scenario in course of proposed project activity lifetime:

At the start of a crediting period, a re-assessment of the baseline scenario due to new or modified  $NO_X$  or  $N_2O$  emission regulations in Brazil, will be executed as follows

**Sub-step 5a**: New or modified NO<sub>X</sub>-emission regulations

If new or modified NO<sub>X</sub> emission regulations are introduced after the project start, determination of the baseline scenario will be re-assessed at the start of a crediting period. Baseline scenario alternatives to be analyzed will include, inter alia:

- Selective Catalytic Reduction (SCR);
- Non-Selective Catalytic Reduction (NSCR);
- Tertiary measures incorporating a selective catalyst for destroying N<sub>2</sub>O and NO<sub>x</sub> emissions;
- Continuation of baseline scenario.

For the determination of the adjusted baseline scenario, the baseline determination process will be applied as stipulated above (Steps 1-5)

**Sub Step 5b:** New or modified N<sub>2</sub>O –regulation

If legal regulations on  $N_2O$  emissions are introduced or changed during the crediting period, the baseline emissions will be adjusted at the time the legislation will be legally implemented.

The methodology is applicable if the procedure to identify the baseline scenario results in that the most likely baseline scenario is the continuation of emitting  $N_2O$  to the atmosphere, without the installation of



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 $N_2O$  destruction or abatement technologies, including technologies that indirectly reduce  $N_2O$  emissions (e.g. NSCR DeNO<sub>X</sub> units).

B.5. Description of how the anthropogenic emissions of GHG by sources are reduced below those that would have occurred in the absence of the registered CDM project activity (assessment and demonstration of additionality):

Fosfertil Nitrous Oxide Abatement Project at Piaçaguera NAP 2 involves the installation of secondary catalysts whose only purpose and effect is the decomposition of nitrous oxide once it is formed.

Following the selected methodology project emissions are determined from  $N_2O$  measurements in the stack gas of the nitric acid plants.

Thus, baseline emissions are calculated from an emission factor measured before the implementation of the project activity, considering that it is physically very difficult to measure  $N_2O$  concentration upstream and downstream of the abatement system accurately.

Then, baseline will be determined by measuring  $N_2O$  baseline emission factor (kg  $N_2O$ /tonne HNO<sub>3</sub>) during a *complete* production campaign, called "initial  $N_2O$  measurement campaign for baseline determination", prior to project implementation.

To ensure that data obtained during such initial campaign are representative of the actual GHG emissions from the source plant, a set of process parameters known to affect  $N_2O$  generation and that are (to some extent) under the control of the plant operator, are monitoring and compared to limits or ranges called "Normal operating conditions".

Normal operating conditions are defined based on plant historical operating conditions, and or plant design data. A range or maximum value for any given parameter has been established considering specific control capabilities of Piaçaguera NAP 2. In order to properly characterize baseline emission rates, operation during such initial campaign is controlled during the specified range (a maximum or range has been established for each parameter). Only those  $N_2O$  measurements taken when the plant is operating within the permitted range will be considered in the calculation of baseline emissions. The level of uncertainty determined for the  $N_2O$  monitoring equipment will be deducted from the baseline emissions factor.

At the moment of presenting this PDD Piaçaguera NAP 2 is beginning their initial campaign for baseline emission factor determination. The baseline campaign began on August, 23<sup>rd</sup>, 2007 and will finish during the middle of February, 2008.

The emissions factor determined from such measurements will be used for crediting of emission reductions.

The additionality of the project activity is demonstrated and assessed using the third version of the "Tool for demonstration and assessment of additionality". We will demonstrate that the baseline scenario is the continuation of the status quo and  $N_2O$  emissions are not reduced by any  $N_2O$  destruction or abatement technology at Piaçaguera NAP 2.

Step 1 of the tool can be avoided since the selection of alternative scenarios was already covered in analysis carried out in section B.4 above.



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### **Step 2. Investment analysis:**

### **Sub-step 2a.** Determine appropriate analysis method:

As catalytic  $N_2O$  destruction facilities generates no financial or economical benefits other than CDM related income, a simple cost analysis is applied.

### Sub-step 2b. – Apply simple cost analysis

Project scenario: No income from any kind of potential product or by-product except CERs are able to pay back investment costs as well as running costs for the installation of the secondary catalyst as no marketable product or by-product exists.

The investment (excluding potential financing costs) consists of the engineering, construction, shipping, installation and commissioning of the secondary catalyst and the measurement equipment. The running costs consist of the regular change of the catalysts as well as personnel costs for the supervision and the measurement equipment.

Baseline scenario: The baseline scenario "The continuation of the current situation" will neither require any additional investments costs nor any additional running costs.

Therefore, the proposed CDM project activity is, without the revenues from the sale of certified emission reductions, obviously less economically and financially attractive than the baseline scenario.

A net present value (NPV) of zero has been chosen to be the relevant financial indicator for the project activity. The NPV is the difference between the sum of the discounted cash flows which are expected from the investment and the amount which is initially invested. This financial indicator is used by most companies including Fosfertil group, to assess the economical value of a project. Unless there is a regulatory constraint, projects are required to have a positive NPV with the discount rate defined by the company's management. Otherwise, they are ruled out. Then, projects are ranked and those with the highest NPVs are selected.

As there is no alternative investment to the project activity that would generate similar services, the NPV is calculated in the following only for the project activity. If the NPV is lower or equals zero the proposed project activity is additional.

The following table shows the net present values (NPV) of the investment in the decomposition device, considering discount rates of 0%, 5%, 10% and 15%.

Net present values (NPV) of the investment in the decomposition facility depending on different discount rates

Discount rate	0%	5%	10%	15%
NPV (R\$)				
	-3,121,666	-2,840,409	-2,613,899	-2,428,995



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**Step 3. Barrier analysis** is not used for demonstrating additionality in this project.

### Step 4. Common practice analysis

The proposed project activity (or any other form of nitrous oxide abatement technology) is not common practice since no similar project at nitric acid plants are identified in Brazil. The nitric acid industry typically releases into the atmosphere the  $N_2O$  generated as a by-product, as it does not have any economic value or toxicity at typical emission levels.  $N_2O$  emissions in the stack gas can be considered the business-as-usual activity and it is spread all over the country. No nitric acid plant in Brazil has a secondary catalyst (or any other type of  $N_2O$  abatement technology) installed, except for Rhodia Poliamida e Especialidades Ltda, at Paulínia, state of Sao Paulo that has installed a secondary catalyst, also as CDM project activity.

Since similar project activities are not observed the proposed project activity is not common practice.

### **Conclusion:**

Currently, there are no national regulations or legal obligations in Brazil concerning  $N_2O$  emissions. It is unlikely that any such limits on  $N_2O$  emissions will be imposed in the near future. In fact, given the cost and complexity of suitable  $N_2O$  destruction and abatement technologies, it is unlikely that a limit would be introduced by Brazil that has ratified the Kyoto Protocol and actively participates in CDM.

Fosfertil is in no need to invest in any  $N_2O$  destruction or abatement technology. Neither are there any national incentives or sectoral policies to promote similar project activities.

Without the sale of the CER's generated by the project activity the NPV and IRR of the project would be negative, no revenue would be generated and the technology would not be installed. The secondary catalyst technology when installed will reduce the Nitrous Oxide emissions by up to 80% below what they would otherwise be without the catalyst technology installed.

The proposed CDM project activity is undoubtedly additional, since it passes all the steps of the "Tool for demonstration and assessment of additionality", approved by the CDM Executive Board. No income from any kind of potential product or by-product except CERs are able to pay back investment costs as well as running costs for the installation of the proposed project activity as no marketable product or by-product exists.

The approval and registration of the project activity as a CDM activity, and the attendant benefits and incentives derived from the project activity, will offset the substantial cost of the catalyst and any plant modifications and will enable the project activity to be undertaken.

Based on the *ex-ante* estimation of N<sub>2</sub>O emission reductions over the first crediting period, it is expected that the income from selling of CERs of the registered CDM project activity is at least as high as the



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investment, financing and running costs. Therefore Fosfertil is willing to finance the project activity under the condition of the registration of the project activity.

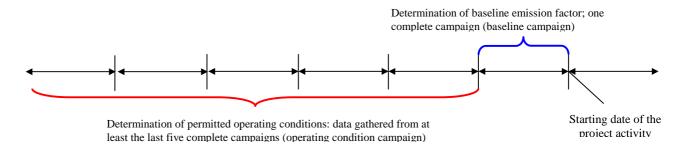
### **B.6.** Emission reductions:

### **B.6.1.** Explanation of methodological choices:

### Baseline emissions procedure

Following AM0034 the baseline shall be established through continuous monitoring of both  $N_2O$  concentration and gas flow volume in the stack of the nitric acid plant for *one complete* campaign prior to project implementation.

The schematic of the procedure is as follows:



# <u>1 - Determination of the permitted operating conditions of the nitric acid plant to avoid overestimation of baseline emissions:</u>

### Oxidation temperature and pressure

For Piaçaguera NAP 2, the range of oxidation temperature and pressure as indicated in the operating manual for the existing equipment is used to determined "permitted operating conditions", due to lack of sufficient historical data.

Technical documents to demonstrate design operating conditions will be available for the validation process of the project activity.

Ammonia gas flow rates and ammonia to air ratio input into the ammonia oxidation reactor

For Piaçaguera NAP 2, the upper limits for ammonia flow and ammonia to air ratio as specified by the ammonia oxidation catalyst manufacturer will be used to determined "permitted operating conditions".



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Ammonia oxidation catalyst design data as per the gauze manufacturer will be available for the validation process of the project activity.

# $\underline{2}$ - Determination of baseline emission factor: measurement procedure for $N_{\underline{2}}O$ concentration and gas volume flow

For the determination of the baseline emission factor  $N_2O$  concentration and gas volume flow will be monitored throughout the baseline campaign. Separate readings for  $N_2O$  concentration and gas flow volume for a defined period of time (e.g. every hour of operation, it provides an average of the measured values for the previous 60 minutes) will be performed. Error readings (e.g. downtime or malfunction) and extreme values will be eliminated from the output data series.

Measurement results can be distorted before and after periods of downtime or malfunction of the monitoring system and can lead to maverick. To eliminate such extremes and to ensure a conservative approach, the following statistical evaluation is to be applied to the complete data series of  $N_2O$  concentration as well as to the data series for gas volume flow. The statistical procedure will be applied to data obtained after eliminating data measured for periods where the plant operated outside the permitted ranges:

- a) Calculate the sample mean (x)
- b) Calculate the sample standard deviation (s)
- c) Calculate the 95% confidence interval (equal to 1.96 times the standard deviation)
- d) Eliminate all data that lie outside the 95% confidence interval
- e) Calculate the new sample mean from the remaining values (volume of stack gas (VSG) and  $N_2O$  concentration of stack gas (NCSG))

Then, the average mass of  $N_2O$  emissions per hour is estimated as product of the NCSG and VSG. The  $N_2O$  emissions per campaign are estimates product of  $N_2O$  emission per hour and the total number of complete hours of operation of the campaign using the following eq. 1 from AM0034:





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$$BE_{BC} = VSG_{BC} \cdot NCSG_{BC} \cdot 10^{-9} \cdot OH_{BC}$$
 (Eq. 1)

where:

 $BE_{BC}$  Total baseline emissions in the baseline measurement period, in, tN<sub>2</sub>O

 $VSG_{BC}$  Mean stack gas volume flow rate in the baseline measurement period, in Nm<sup>3</sup>/h

 $NCSG_{BC}$  Mean concentration of N<sub>2</sub>O in the stack gas in the baseline measurement period, in mg

 $N_2O/Nm^3$ 

 $OH_{BC}$  Number of operating hours in the baseline measurement period, in h

The plant specific baseline emissions factor representing the average  $N_2O$  emissions per tonne of nitric acid over *one full campaign* is derived by dividing the total mass of  $N_2O$  emissions by the total output of 100% concentrated nitric acid for that period for baseline emission factor determination.

Following AM0034, the  $N_2O$  emission factor per tonne of nitric acid produced in the baseline period  $(EF_{BL})$  will be reduced by the estimated percentage error (UNC):

$$EF_{BL} = \frac{BE_{BC}}{NAP_{RC}} (1 - \frac{UNC}{100})$$
 (Eq. 2)

where:

 $EF_{BL}$  Baseline emission factor, in  $tN_2O/tHNO_3$ 

 $NAP_{BC}$  Nitric acid production during the baseline campaign, in, tHNO<sub>3</sub>

UNC Overall measurement uncertainty of the monitoring system, in %, calculated as the

combined uncertainty of the applied monitoring equipment

### <u>Impact of regulations</u>

Should  $N_2O$  emissions regulations that apply to nitric acid plants be introduced in the Brazil or jurisdiction covering the location of nitric acid plants, such regulations shall be compared to the calculated baseline emission factor ( $EF_{BL}$ ), regardless of whether the regulatory level is expressed as:

- An absolute cap on the total volume of N<sub>2</sub>O emissions for a set period;
- A relative limit on N<sub>2</sub>O emissions expressed as a quantity per unit of output; or
- A threshold value for specific N<sub>2</sub>O mass flow in the stack;

In this case, a corresponding plant-specific emissions factor cap (max. allowed tN<sub>2</sub>O/tHNO<sub>3</sub>) is to be derived from the regulatory level. If the regulatory limit is lower than the baseline factor determined for the project activity, the regulatory limit shall become as the new baseline emission factor, that is.

If  $EF_{BL} > EF_{reg}$ , then  $EF_{BL} = EF_{reg}$  for all the calculations.



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### Composition of the ammonia oxidation catalyst

The composition of the ammonia oxidation catalyst used for the baseline campaign and after the implementation of the project are identical to that used in the campaigns for setting the operating conditions (previous five campaigns), then there shall be no limitations on  $N_2O$  baseline emissions.

### Campaign Length

In order to take into account the variations in campaign length and its influence on  $N_2O$  emission levels, the historic campaign lengths and the baseline campaign length are to be determined and compared to the project campaign length. Campaign length is defined as the total number of metric tonnes of nitric acid at 100% concentration produced with one set of gauzes.

### Historic Campaign Length

The average historic campaign length ( $CL_{normal}$ ) defined as the average campaign length for the historic campaigns used to define operating condition (the previous five campaigns), will be used as a cap on the length of the baseline campaign.

Historical data and statistical analysis to determine "historic campaign length" will be available for the validation process of the project activity.

If baseline campaign length ( $CL_{BL}$ ) is lower or equal than  $CL_{normal}$ , all N<sub>2</sub>O values measured during the baseline campaign can be used for the calculation of  $EF_{BL}$  (subject to the elimination of data that was monitored during times where the plant was operating outside of the "permitted range").

If baseline campaign length ( $CL_{BL}$ ) is higher than  $CL_{normal}$ , all N<sub>2</sub>O values measured beyond the length of  $CL_{normal}$  during the production of the quantity of nitric acid (i.e. the final tonnes produced) will be eliminated from the calculation of  $EF_{BL}$ ."

Parameters to be monitored for composition of the catalyst are as follows:

GS<sub>normal</sub> Gauze supplier for the operation condition campaigns

GS<sub>BC</sub> Gauze supplier for baseline campaign

GS<sub>project</sub> Gauze supplier for the project campaign

GC<sub>normal</sub> Gauze composition for the operation condition campaigns

GC<sub>BC</sub> Gauze composition for baseline campaign

GC<sub>project</sub> Gauze composition for the project campaign

### Project emission procedure

Actual project emissions will be determined during the project activity from continuous measurements of  $N_2O$  concentration and total flow rate in the stack gas of the nitric acid plant.

Project measurements are subjected to exactly the same procedure as the baseline measurements in order to be coherent.



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### Estimation of campaign-specific project emissions

The monitoring system will provide separate reading for  $N_2O$  concentration and gas flow for a define period of time (e.g. every hour of operation, i.e. an average of the measuring values of the past 60 minutes). Error readings (e.g. downtime or malfunction) and extreme values are eliminated from the output data series. Next, the same statistical evaluation that was applied to the baseline data series has to be applied to the project data series:

- a) calculate the sample mean (x)
- b) calculate the sample standard deviation (s)
- c) calculate the 95% confidence interval (equal to 1.96 times the standard deviation)
- d) eliminate all data that lie outside the 95% confidence interval
- e) calculate the new sample mean from the remaining values

The mean values of N<sub>2</sub>O concentration and total flow rate are used in the following formula (Eq. 3 from AM0034) to calculate project emissions:

$$PE_n = VSG_n \cdot NCSG_n \cdot 10^{-9} \cdot OH_n \tag{Eq. 3}$$

where:

 $PE_n$  Total Project emissions of the nth campaign, in tN<sub>2</sub>O

 $VSG_n$  Mean stack gas volume flow rate for the nth project campaign, in Nm<sup>3</sup>/h

 $NCSG_n$  Mean concentration of  $N_2O$  in the stack gas for the project campaign, in mg  $N_2O/Nm^3$ 

 $OH_n$  Number of operating hours in the project campaign, in h

### Derivation of a moving average emission factor

In order to take into account possible long-term emissions trends over the duration of the project activity and to take a conservative approach a moving average emission factor is estimated as follows:

Step1: estimate campaign specific emissions factor for each campaign during the project's crediting period by dividing the total mass of  $N_2O$  emissions during that campaign by the total production of 100% concentrated nitric acid during that same campaign.

For example, for the *nth* campaign the campaign specific emission factor would be:



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$$EF_n = \frac{PE_n}{NAP_n}$$
 (Eq. 4)

where:

 $EF_n$  Emission factor calculated for the *nth* campaign, in kg  $N_2O/ton\ HNO_3$ 

 $PE_n$  Total Project emissions of the *nth* campaign, in tN<sub>2</sub>O

 $NAP_n$  Nitric acid production in the *nth* campaign, in ton 100% HNO<sub>3</sub>

Step 2: estimate a moving average emissions factor calculated at the end of the *nth* project campaign as follows:

$$EF_{ma,n} = \frac{\sum_{n} EF_{n}}{n}$$
 (Eq. 5)

This process will be repeated for each campaign such that a moving average,  $EF_{ma,n}$  is established over time, becoming more representative and precise with each additional campaign.

To calculate the total emission reductions achieved in the *nth* campaign, the higher of the two values  $EF_{ma,n}$  and  $EF_{,n}$  shall be applied as the emission factor relevant for that particular campaign (EF<sub>p</sub>).

If 
$$EF_{ma,n} > EF_{,n}$$
, then  $EF_{,p} = EF_{ma,n}$   
If  $EF_{ma,n} < EF_{,n}$ , then  $EF_{,p} = EF_{,n}$  (Eq. 6)

### Minimum project emission factor

A campaign-specific emissions factor shall be used to cap any potential long-term trend towards decreasing  $N_2O$  emissions that may result from a potential built up of platinum deposits. After the first ten campaigns of the crediting period of the project, the lowest  $EF_n$  observed during those campaigns will be adopted as a minimum ( $EF_{min}$ ). If any of the later project campaigns results in an  $EF_n$  that is lower than  $EF_{min}$ , the calculation of the emission reductions for that particular campaign shall used  $EF_{min}$  and not  $EF_n$ .

### Project Campaign Length

### a. Longer Project Campaign

If the length of each individual project campaign  $CL_n$  is longer than or equal to the average historic campaign length  $CL_{normal}$ , then all N<sub>2</sub>O values measured during the baseline campaign can be used for the calculation of  $EF_n$  (subject to the elimination of data from the Ammonia/Air analysis).

### b. Shorter Project Campaign





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If  $CL_n < CL_{normal}$ , recalculate  $EF_{BL}$  by eliminating those N<sub>2</sub>O values that were obtained during the production of tonnes of nitric acid beyond the  $CL_n$  (i.e. the last tonnes produced) from the calculation of  $EF_n$ .

### Leakage procedure

No leakage calculation is required.

#### **Emission reduction calculation**

The emission reductions of the project activity, ER, expressed in tonnes of  $CO_2$  equivalent per year ( $tCO_2e/yr$ ), are given by Eq. 7 (Eq. 7 from AM0034):

$$ER_n = (EF_{RL} - EF_p) \cdot NAP_n \cdot GWP_{N,O}$$
 (Eq. 7)

where

 $ER_n$  Emission reductions for the *nth* campaign,  $tCO_2e$ 

 $EF_{BL}$  Baseline emission factor, in  $tN_2O/tHNO_3$  $EF_p$  Project emission factor, in  $tN_2O/tHNO_3$ 

NAP Nitric acid production during the *nth* campaign of the project activity, in, tHNO<sub>3</sub>

 $GWP_{N,O}$  Global warming potential, of N<sub>2</sub>O set as 310 tCO<sub>2</sub>e/tN<sub>2</sub>O for the 1<sup>st</sup> commitment period

Note. The nitric acid production used to calculate emission reduction should not exceed the design capacity (nameplate) of the nitric acid plant.

Documentation to prove design capacity (nameplate) of the nitric acid plant will be available for the validation process of the project activity.<sup>5</sup>

### **B.6.2.** Data and parameters that are available at validation:

(Copy this table for each data and parameter)

Data / Parameter:	Normal Operating Temperature, OT <sub>normal</sub> (range of temperature)
Data unit:	$^{\circ}$ C
Description:	Range of oxidation temperature of the ammonia reactor
Source of data used:	Calculated from Plant design data
Value applied:	850°C- 920°C

<sup>&</sup>lt;sup>5</sup> By nameplate (design) implies the total yearly capacity (considering 365 days of operation per year) as per the documentation of the plant technology provider (such as the Operation Manual). If the plant has been modified to increase production, and such de-bottleneck or expansion projects were completed before December 2005, then the new capacity is considered nameplate, provided proper documentation of the projects is available (such as, but not limited to: properly dated engineering plans or blueprints, engineering, materials and/or equipment expenses, or third party construction services, etc.)



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Justification of the	At Piaçaguera NAP 2 complete historical registers of oxidation temperature
choice of data or	(previous 5 campaigns) are not available, then plant design data was used to
description of	determine normal oxidation temperature
measurement methods	
and procedures	
actually applied:	
Any comment:	None

Data / Parameter:	Normal Operating Pressure, OP <sub>normal</sub> (range of pressure)
Data unit:	Pa (abs)
Description:	Range of oxidation pressure of the ammonia reactor.
Source of data used:	Calculated from Plant design data
Value applied:	560,000 Pa abs - 640,000 Pa abs
Justification of the	At Piaçaguera NAP 2 complete historical registers of oxidation pressure
choice of data or	(previous 5 campaigns) are not available, then plant design data was used to
description of	determine normal oxidation pressure
measurement methods	
and procedures	
actually applied:	
Any comment:	None

Data / Parameter:	Maximum Ammonia Flow Rate, AFR <sub>max</sub>
Data unit:	kg NH <sub>3</sub> /hour
Description:	Ammonia flow rate to the ammonia oxidation reactor.
Source of data used:	Defined considering option b, as specified by the ammonia oxidation catalyst
	manufacturer, Umicore.
Value applied:	4,000 kg NH <sub>3</sub> /hour
Justification of the	At Piaçaguera NAP 2, complete historical registers for ammonia flow to reactor
choice of data or	(previous 5 campaigns) are unavailable; then, maximum ammonia load as
description of	specified by the primary catalyst manufacturer was used to determine
measurement methods	maximum ammonia flow rate.
and procedures	
actually applied:	
Any comment:	None

Data / Parameter:	Maximum Ammonia to Air Flow Rate, AIFR <sub>max</sub>
Data unit:	Kg NH <sub>3</sub> /kg air
Description:	Ammonia to air flow rate to the ammonia oxidation reactor.
Source of data used:	Defined considering option b, as specified by the ammonia oxidation catalyst
	manufacturer, Umicore.
Value applied:	0.0746 kg NH <sub>3</sub> /kg air
Justification of the	At Piaçaguera Plant 2, complete historical registers for ammonia flow rate to
choice of data or	reactor (previous 5 campaigns) are unavailable; then, maximum load as





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description of	specified by the primary catalyst manufacturer was used to determine
measurement methods	maximum ammonia to air flow rate.
and procedures	
actually applied:	
Any comment:	None

Data / Parameter:	Normal Campaign Length, CL <sub>normal</sub>
Data unit:	ton 100% HNO <sub>3</sub>
Description:	Campaign length is defined as the total number of metric tonnes of nitric acid at
	100% concentration produced with one set of gauzes.
Source of data used:	Calculated form historical process data.
Value applied:	49,734 ton 100% HNO <sub>3</sub>
Justification of the	Calculated as described (above), from historical data
choice of data or	Daily production was measured by using a volume flow meter (Magnetic head
description of	principle), and correcting by the average of several (3 to 5) concentration
measurement methods	checkups performed in analytical lab
and procedures	
actually applied:	
Any comment:	None

Data / Parameter:	Normal Gauze Supplier, GS <sub>normal</sub>
Data unit:	Umicore
Description:	Gauze supplier during operating condition campaigns (the previous five campaigns).
Source of data used:	From historical process data
Value applied:	Umicore
Justification of the choice of data or description of measurement methods and procedures actually applied:	Umicore supplies primary catalyst package to Fosfertil on a contract basis for commercial/economic reasons
Any comment:	None

Data / Parameter:	Normal Gauze Composition, GC <sub>normal</sub>
Data unit:	%
Description:	Gauze composition for the operation condition campaigns (the previous five campaigns).  Fosfertil Piaçaguera Plant 2 has 3 woven gauzes and 6 knitted gauzes
Source of data used:	From historical process data.
Value applied:	Pt 95%, Rh 5%
Justification of the	Current gauze composition delivers acceptable performance (as per contractual
choice of data or	basis considering commercial/economic issues







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description of	
measurement methods	
and procedures	
actually applied:	
Any comment:	None

### **B.6.3** Ex-ante calculation of emission reductions:

For completing this PDD with the estimation of project emissions the following assumptions are used:

- Nitric acid production is assumed to be constant, so that project emissions do not vary from year
  to year. Nitric Acid production is estimated as the daily average production multiplied by the
  operational factor (360 days per year).
- An N<sub>2</sub>O emission factor (EF<sub>BL</sub>) calculated from monitored data available at the moment of submitting this PDD. The final baseline emission factor will be calculated after the completion of baseline campaign measurements
- The technology provider, indicate that the estimated reduction efficiency to be achieved as a consequence of project implementation is at least 80%. Then, in order to present estimative values in this PDD, we consider  $N_2O$  concentration in the stack gas during the project campaign equal to 20% of the  $N_2O$  concentration in the stack gas during the baseline campaign (NCSG =  $0.20 * NSCG_{BC}$ )

Then, ex-ante estimations of emission reduction are determined using the following equations:

$$BE_{BC} = VSG_{BC} \cdot NCSG_{BC} \cdot 10^{-9} \cdot OH_{BC}$$
 (Eq. 8)

$$BE_{BC} = 34,486 \cdot 2,525 \cdot 168 \cdot 10^{-9} = 14.63 \, tN_2O$$
 (Eq. 9)

$$EF_{BL} = \frac{BE_{BC}}{NAP_{BC}} (1 - \frac{UNC}{100})$$
 (Eq. 10)

$$EF_{BL} = \frac{14.63}{1,864} \cdot (1 - \frac{5.79}{100}) = 0.00739tN_2O/tHNO_3$$
 (Eq. 11)

$$PE_n = VSG \cdot NCSG \cdot 10^{-9} \cdot OH$$
 (Eq. 12)

$$PE_n = 34,486 \cdot 0.20 \cdot 2,525 \cdot 10^{-9} \cdot 4,512 = 78.57 \, tN_2O$$
 (Eq. 13)

$$EF_p = \frac{PE_n}{NAP_n} \tag{Eq. 14}$$





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$$EF_p = \frac{78.57}{49.734} = 0.00158 \, tN_2 O / tHNO_3$$
 (Eq. 15)

Then.

$$ER_n = (EF_{RL} - EF_n) \cdot NAP \cdot GWP_{N_2O}$$
 (Eq. 16)

$$ER_n = (0.00739 - 0.00158) \cdot 95,400 \cdot 310 = 171,931 ton CO_2 e / year$$
 (Eq. 17)

### Where

NAP

 $GWP_{N2O}$ 

 $BE_{BC}$ Total baseline emissions in the baseline measurement period, in, tN<sub>2</sub>O Mean stack gas volume flow rate in the baseline measurement period, in Nm<sup>3</sup>/h  $VSG_{BC}$ Mean concentration of N<sub>2</sub>O in the stack gas in the baseline measurement period, in  $NCSG_{BC}$  $mg N_2O/Nm^3$  $OH_{BC}$ Number of operating hours in the baseline measurement period, in h Baseline emission factor, in tN<sub>2</sub>O/ tHNO<sub>3</sub>  $EF_{BL}$  $NAP_{BC}$ Nitric acid production during the baseline campaign, in, tHNO<sub>3</sub> Overall uncertainty of the AMS, % UNC  $PE_n$ Estimated N<sub>2</sub>O emission for the project campaign, tN<sub>2</sub>O **VSG** Estimated mean stack gas volume flow rate for the project campaign, in Nm<sup>3</sup>/h Estimated mean concentration of N<sub>2</sub>O in the stack gas for the project campaign, in NCSG  $mg N_2O/Nm^3$ OHEstimated number of operating hours in the project campaign, in h Estimated project emission factor, in tN<sub>2</sub>O/tHNO<sub>3</sub>  $EF_{p}$  $NAP_n$ Nitric acid production for the project campaign, tHNO<sub>3</sub>  $ER_n$ Emission reductions for the *nth* campaign, tCO<sub>2</sub>e

Nitric acid production during year y, in, tHNO<sub>3</sub>/year

The assumptions parameters are specified in the following table:

<b>Estimated values</b>	Fosfertil Piaçaguera NAP 2
NAP, t HNO₃/yr	95,400
$OH_{,}h$	4,512
$GWP_{N_2O} \ tCO_2e/tN_2O$	310

Global Warming Potential of N<sub>2</sub>O set as 310 tCO<sub>2</sub>e/tN<sub>2</sub>O for the 1<sup>st</sup> commitment period

**Note:** In order to follow the calculations see Spreadsheet "Baseline Campaign-Fosfertil Piaçaguera NAP 2-25Sept2007".xls"

### **B.6.4** Summary of the ex-ante estimation of emission reductions:







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The ex-ante estimations of project emission reductions are summarized in the table below:

Years	Estimation of project activity emissions (tonnes of CO <sub>2</sub> e)	Estimation of baseline emissions (tonnes of CO <sub>2</sub> e)	Estimation of leakage (tonnes of CO <sub>2</sub> e)	Estimation of overall emission reductions (tonnes of CO <sub>2</sub> e)
2008 <sup>6</sup>	40,882	191,321	-	150,440
2009	46,722	218,653	-	171,931
2010	46,722	218,653	-	171,931
2011	46,722	218,653	-	171,931
2012	46,722	218,653	-	171,931
2013	46,722	218,653	-	171,931
2014	46,722	218,653	-	171,931
2015 <sup>7</sup>	5,840	27,332	-	21,491
Total (tones of CO <sub>2</sub> e)	327,054	1,530,571	-	1,203,571

<sup>6</sup> Note: Year 2008 includes ten months and a half, from Middle of February to December Note: Year 2015 includes one month and a half, from January to middle of February

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# B.7 Application of the monitoring methodology and description of the monitoring plan:

# **B.7.1** Data and parameters monitored:

(Copy this table for each data and parameter)

Data / Parameter:	Baseline Volume Flow in the Stack Gas, VSG <sub>BC</sub>
Data unit:	Nm <sup>3</sup> / hour
Description:	Mean gas volume flow rate in the stack gas during baseline campaign
Source of data to be used:	AMS (Flow meter) at Fosfertil Piaçaguera NAP 2
Value of data applied	34,486 Nm <sup>3</sup> / hour
for the purpose of	
calculating expected	
emission reductions in	
section B.5	
Description of	Stack flow is measured by ANNUBAR device (multiple pressure differential
measurement methods	principle) with automatically compensates for actual stack pressure and
and procedures to be applied:	temperature in order to normalize output data
QA/QC procedures to	Regular calibrations according to vendor specifications and recognised
be applied:	industry standards (EN 14181) as reference method. Staff will be trained in
	monitoring procedures and a reliable technical support infrastructure will set
	up.
Any comment:	Measured during a complete campaign before project implementation to
	properly characterize baseline emissions factor.
	Recorded every two seconds

Data / Parameter:	Baseline Temperature of the Stack Gas, TSG <sub>BC</sub>
Data unit:	°C
Description:	Temperature of the gas in the stack gas during baseline campaign
Source of data to be	AMS (Flow meter).
used:	
Value of data applied	Not applicable. We do not use this parameter to estimate expected emission
for the purpose of	reduction.
calculating expected	
emission reductions in	
section B.5	
Description of	Stack flow is measured by ANNUBAR device (multiple pressure differential
measurement methods	principle) with automatically compensates for actual stack pressure and
and procedures to be	temperature in order to normalize output data.
applied:	
QA/QC procedures to	Regular calibrations according to vendor specifications and recognised
be applied:	industry standards. Staff will be trained in monitoring procedures and a reliable
	technical support infrastructure will be set up.





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Any comment:	Measured during a complete campaign before project implementation to
	properly characterize baseline emissions factor.
	Recorded every two seconds

Data / Parameter:	Baseline Pressure of the Stack Gas, PSG <sub>BC</sub>
Data unit:	$\text{Kg/cm}^2$
Description:	Pressure in the stack gas during baseline campaign
Source of data to be	AMS (Flow meter)
used:	
Value of data applied	Not applicable. We do not use this parameter to estimate expected emission
for the purpose of	reduction.
calculating expected	
emission reductions in	
section B.5	
Description of	Stack flow is measured by ANNUBAR device (multiple pressure differential
measurement methods	principle) with automatically compensates for actual stack pressure and
and procedures to be	temperature in order to normalize output data.
applied:	
QA/QC procedures to	Regular calibrations according to vendor specifications and recognised
be applied:	industry standards. Staff will be trained in monitoring procedures and a reliable
	technical support infrastructure will be set up.
Any comment:	Measured during a complete campaign before project implementation to
	properly characterize baseline emissions factor.
	Recorded every two seconds

Data / Parameter:	Baseline N <sub>2</sub> O Concentration in the Stack Gas, NCSG <sub>BC</sub>
Data unit:	mg N <sub>2</sub> O/ Nm <sup>3</sup> (converted from ppm if necessary)
Description:	Mean concentration of N <sub>2</sub> O in the stack gas for the baseline campaign
Source of data to be	AMS (Infrared gas analyzer) at Fosfertil Piaçaguera NAP 2.
used:	
Value of data applied	$2,525 \text{ mg N}_2\text{O/Nm}^3$
for the purpose of	
calculating expected	
emission reductions in	
section B.5	
Description of	N <sub>2</sub> O concentration is measured by on-line analyzer (Non Dispersive Infra Red
measurement methods	principle). A gas stream is continuously drawn from the stack by the sampling
and procedures to be	system under proper conditions (line is heat traced to avoid condensation), and
applied:	driven to the infrared cell. The device is set up to measure concentration and
	record the output electronically every 2 seconds.
QA/QC procedures to	Regular calibrations according to vendor specifications and recognised
be applied:	industry standards (EN 14181) as reference method. Staff will be trained in
	monitoring procedures and a reliable technical support infrastructure will set
	up.





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Any comment:	Measured during a complete campaign before project implementation to
	properly characterize baseline emissions factor.
	Recorded every two seconds

Data / Parameter:	Baseline Operating Hours, OH <sub>BC</sub>
Data unit:	Hours
Description:	Total operating hours for the baseline campaign
Source of data to be	Process control system at Fosfertil Piaçaguera NAP 2.
used:	
Value of data applied	168 hours
for the purpose of	
calculating expected	
emission reductions in	
section B.5	
Description of	Operated hours are measured by the consumption of ammonia registered. The
measurement methods	unit begins the operation with the ammonia introduction in the NO converter
and procedures to be	and, the end of the operation is characterized with the cut of ammonia flow to
applied:	this reactor.
QA/QC procedures to	Critical instruments are calibrated on a routinely basis according to the quality
be applied:	assurance system of the plant.
Any comment:	Measured daily during a complete campaign before project implementation to
	properly characterize baseline emissions factor.

Data / Parameter:	Uncertainty of the monitoring system, UNC
Data unit:	%
Description:	Overall uncertainty of the monitoring system, calculated as the combined
	uncertainty of the applied monitoring equipment
Source of data to be	The infrared analyzer supplier will issue a manufacturer's performance
used:	declaration. The declaration specifies the uncertainty level of the instrument.
	Uncertainty from the flow measuring device will also be estimated and
	accounted for using proper statistical methods, to determine the overall
	uncertainty of the AMS (Automated Measuring System).
Value of data applied	5.79%
for the purpose of	
calculating expected	
emission reductions in	
section B.5	
Description of	The overall uncertainty was determined summarizing the uncertainty of the
measurement methods	flow meter and the uncertainty of the $N_2O$ concentration measurements.
and procedures to be	
applied:	
QA/QC procedures to	No QA/QC procedure is needed.
be applied:	
Any comment:	Calculated once



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Data / Parameter:	Nitric Acid Production, NAP <sub>BC</sub>
Data unit:	ton 100% HNO <sub>3</sub>
Description:	Total nitric acid production for the baseline campaign
Source of data to be	Production logs of Fosfertil Piçaguera NAP 2.
used:	
Value of data applied	Nitric acid production is assumed to be constant, so that project emissions do
for the purpose of	not vary from year to year. Nitric Acid production is estimated as the daily
calculating expected	average production multiplied by the operational factor (360 days per year).
emission reductions in	The value of the nitric acid production used for the calculation of expected
section B.5	emission reductions is: 95,400 ton HNO <sub>3</sub> /year (based on historical data)
Description of	Daily production is measured by using a volume flow meter (Magnetic head
measurement methods	principle), and correcting by the average of several (3 to 5) concentration
and procedures to be	checkups performed in analytical lab
applied:	
QA/QC procedures to	Mass balance at the end-product storage tanks is performed on a monthly basis;
be applied:	cross checking of production from mass balance with direct flow measurement
	is performed routinely. Critical instruments are calibrated on a routinely basis
	according to the quality assurance system of the plant.
Any comment:	Measured daily during a complete campaign before project implementation to
	properly characterize baseline emissions factor.

Data / Parameter:	Baseline Emission Factor, EF <sub>BL</sub>
Data unit:	ton N2O / ton 100% HNO3
Description:	Baseline emission factor is calculated from monitored data for the baseline
	campaign
Source of data to be	Calculated from monitored data.
used:	
Value of data applied	For the purpose of calculating expected emission reductions, an N <sub>2</sub> O emission
for the purpose of	factor (EF <sub>BL</sub> ) calculated from monitored data available at the moment of
calculating expected	submitting this PDD is used. The final baseline emission factor will be
emission reductions in	calculated after the completion of the baseline campaign measurements. It is
section B.5	expected that the final emission factor will be higher than the emission factor
	used in this PDD since the emission factor increases during the campaign. The
	N <sub>2</sub> O emission factor used for ex ante calculation is:
	$0.00739 \text{ ton } N_2O/\text{ ton } 100\% \text{ HNO}_3$
Description of	Calculated from monitored data.
measurement methods	
and procedures to be	
applied:	
QA/QC procedures to	No QA/QC procedure is needed.
be applied:	





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Any comment:	Baseline emission factor per unit of nitric acid produced will be calculated based on measurements of the nitric acid production, stack gas flow rate, $N_2O$ concentration, and the operating hours. All parameters will be measured during a complete campaign before project implementation to properly characterize baseline emissions factor. Calculated once at the end of the baseline campaign
	Calculated once at the end of the baseful campaign

Data / Parameter:	Baseline Oxidation Temperature, OT <sub>BC</sub>
Data unit:	$^{\circ}$ C
Description:	Oxidation temperature of the ammonia reactor for the baseline campaign
Source of data to be	Distributed Control System of Fosfertil Piaçaguera NAP 2.
used:	
Value of data applied	The permitted range has been established using plant design data (850°C-
for the purpose of	920°C).
calculating expected	Any N <sub>2</sub> O baseline data that is measured during hours where the oxidation
emission reductions in	temperature is outside the permitted range will be eliminated from the
section B.5	calculation of the baseline emissions factor.
Description of	Reactor temperature is measured by four thermocouples installed through the
measurement methods	reactor wall, near the oxidation catalyst; the signal from such device is
and procedures to be	acquired by the Distributed Control System and stored electronically at a given
applied:	time interval.
QA/QC procedures to	New thermocouple units are installed immediately after failure reading (open
be applied:	signal).
Any comment:	Monitored during the initial campaign for baseline emission factor
	determination, in order to avoid manipulations that could increase baseline
	$N_2O$ formation.
	Measured every hour

Data / Parameter:	Baseline Oxidation Pressure, OP <sub>BC</sub>
Data unit:	Pa abs
Description:	Oxidation pressure of the ammonia reactor for the baseline campaign
Source of data to be	Distributed Control System of Fosfertil Piaçaguera NAP 2.
used:	
Value of data applied	The permitted range has been established using plant design data (560,000 Pa
for the purpose of	abs - 640,000 Pa abs)
calculating expected	Any N <sub>2</sub> O baseline data that is measured during hours where the oxidation
emission reductions in	pressure is outside the permitted range will be eliminated from the calculation
section B.5	of the baseline emissions factor.
Description of	Oxidation pressure is tracked by measuring device installed locally near
measurement methods	oxidation reactor. Currently, data is acquired manually on operation log book.
and procedures to be	A measuring device with electronic signal transmission will be installed before
applied:	baseline campaign; the signal will be acquired by the Distributed Control
	System and stored electronically at a given time interval.
QA/QC procedures to	Critical instruments are calibrated on a routinely basis according to the quality



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be applied:	assurance system of the plant
Any comment:	Monitored during the initial campaign for baseline emission factor
	determination, in order to avoid manipulations that could increase baseline
	$N_2O$ formation.
	Measured every hour

Data / Parameter:	Baseline Ammonia Flow Rate, AFR <sub>BC</sub>
Data unit:	Kg NH <sub>3</sub> /hour
Description:	Ammonia flow rate to the ammonia oxidation reactor for the baseline
	campaign.
Source of data to be	Distributed Control System of Fosfertil Piaçaguera NAP 2.
used:	
Value of data applied	The permitted range has been established using plant design data (4,000 kg
for the purpose of	NH <sub>3</sub> /hour)
calculating expected	Any N <sub>2</sub> O baseline data that is measured during hours where the ammonia flow
emission reductions in	rate is outside the permitted range will be eliminated from the calculation of
section B.5	the baseline emissions factor.
Description of	Ammonia flow to oxidation reactor is tracked by a flow measuring device (V-
measurement methods	Cone principle); the signal from the device is acquired by the Distributed
and procedures to be	Control System and stored electronically at a given time intervals. The System
applied:	is compensated by pressure and temperature.
QA/QC procedures to	Overall mass balances of ammonia to the plant are performed on a monthly
be applied:	basis to cross check flows and overall conversion efficiency. Critical
	instruments are calibrated on a routinely basis according to the quality
	assurance system of the plant.
Any comment:	Monitored continuously during the initial campaign for baseline emission
	factor determination, in order to avoid manipulations that could increase
	baseline N <sub>2</sub> O formation.





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Data / Parameter:	Baseline Ammonia to Air Flow Rate, AIFR <sub>BC</sub>
Data unit:	(Kg NH <sub>3</sub> / Kg Air)
Description:	Ammonia to air flow rate to the ammonia oxidation reactor for the baseline
	campaign.
Source of data to be	Distributed Control System of Fosfertil Piaçaguera NAP 2.
used:	
Value of data applied	The permitted range has been established using plant design data (0.0746 kg
for the purpose of	NH <sub>3</sub> /kg air)
calculating expected	Any N <sub>2</sub> O baseline data that is measured during hours where the ammonia to air
emission reductions in	flow rate is outside the permitted range will be eliminated from the calculation
section B.5	of the baseline emissions factor.
Description of	Ammonia flow to oxidation reactor is tracked by a flow measuring device (V-
measurement methods	Cone principle), and the air flow to oxidation reactor is tracked by a flow
and procedures to be	measuring device (Venturi principle); signals from devices are acquired by the
applied:	Distributed Control System and stored electronically at a given time intervals.
	The System is compensated by pressure and temperature.
QA/QC procedures to	Overall mass balances of ammonia to the plant are performed on a routine
be applied:	basis to cross check flows and overall conversion efficiency. Critical
	instruments are calibrated on a routinely basis according to the quality
	assurance system of the plant
Any comment:	Monitored during the initial campaign for baseline emission factor
	determination, in order to avoid manipulations that could increase baseline
	$N_2O$ formation.
	Measured every hour

Data / Parameter:	Baseline Campaign Length, CL <sub>BL</sub>
Data unit:	ton 100% HNO <sub>3</sub>
Description:	Campaign length is defined as the total number of metric tonnes of nitric acid
	at 100% concentration produced with one set of gauzes. (see baseline nitric
	acid production, NAP <sub>BC</sub> )
Source of data to be	Plant production log book
used:	
Value of data applied	The normal campaign length has been set as 49,734 ton 100% HNO <sub>3</sub>
for the purpose of	If production at a given campaign is lower or equal than normal ( $CL_{Normal}$ ), then
calculating expected	the baseline is recalculated by ignoring the data generated after production
emission reductions in	exceeds normal campaign length.
section B.5	
Description of	Daily production is measured by using a flow meter (Magnetic head principle),
measurement methods	and correcting by the average of several (3 to 5) concentration checkups
and procedures to be	performed in analytical lab.
applied:	
QA/QC procedures to	Cross checking of production from mass balance with direct flow measurement
be applied:	is performed routinely. Critical instruments are calibrated on a routinely basis
	according to the quality assurance system of the plant.



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Any comment:	Measured once
--------------	---------------

Data / Parameter:	Baseline Gauze Supplier, GS <sub>BC</sub>
Data unit:	Umicore
Description:	Gauze supplier for the baseline campaign
Source of data to be	Nitric plant procurement office
used:	
Value of data applied	Umicore
for the purpose of	
calculating expected	
emission reductions in	
section B.5	
Description of	Cover of supply contract for gauzes for baseline campaign, or equivalent
measurement methods	document to prove commercial transaction.
and procedures to be	
applied:	
QA/QC procedures to	None
be applied:	
Any comment:	Recording once

Data / Parameter:	Baseline Gauze Composition, GC <sub>BC</sub>
Data unit:	% precious metals (Platinum, Rhodium)
Description:	Gauze composition for the baseline campaign
Source of data to be	Nitric plant procurement office and/or gauze Supplier technical service
used:	department.
Value of data applied	Pt 95%, Rh 5%
for the purpose of	
calculating expected	
emission reductions in	
section B.5	
Description of	Section of supply contract for gauzes that specifies the technical characteristics
measurement methods	agreed during baseline campaign. If necessary, additional data could be
and procedures to be	requested to supplier's technical service office in order to provide complete
applied:	technical profile of gauzes.
QA/QC procedures to	None
be applied:	
Any comment:	Recording once



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Data / Parameter:	Project Volume Flow in the Stack Gas, VSG <sub>project</sub>
Data unit:	N m <sup>3</sup> / hour
Description:	Volume flow rate in the stack gas for the project campaign
Source of data to be	AMS (Flow meter) at Fosfertil Piaçaguera NAP 2.
used:	
Value of data applied	34,486 Nm <sup>3</sup> / hour
for the purpose of	
calculating expected	
emission reductions in	
section B.5	
Description of	Stack flow is measured by ANNUBAR device (multiple pressure differential
measurement methods	principle) with automatically compensates for actual stack pressure and
and procedures to be	temperature in order to normalize output data.
applied:	
QA/QC procedures to	Regular calibrations according to vendor specifications and recognised
be applied:	industry standards (EN 14181). Staff will be trained in monitoring procedures
	and a reliable technical support infrastructure will set up.
Any comment:	Measured during the complete lifetime of the project activity.
	Recorded every two seconds.

Data / Parameter:	Project Temperature of the Stack Gas, TSG <sub>project</sub>
Data unit:	$^{\circ}\mathrm{C}$
Description:	Temperature of the gas in the stack gas during project campaign
Source of data to be	AMS (Flow meter).
used:	
Value of data applied	Not applicable. We do not use this parameter to estimate expected emission
for the purpose of	reduction.
calculating expected	
emission reductions in	
section B.5	
Description of	Stack flow is measured by ANNUBAR device (multiple pressure differential
measurement methods	principle) with automatically compensates for actual stack pressure and
and procedures to be	temperature in order to normalize output data
applied:	
QA/QC procedures to	Regular calibrations according to vendor specifications and recognised
be applied:	industry standards. Staff will be trained in monitoring procedures and a reliable
	technical support infrastructure will be set up.
Any comment:	Measured during the complete lifetime of the project activity.
	Recorded every two seconds.





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Data / Parameter:	Project Pressure of the Stack Gas, PSG <sub>project</sub>
Data unit:	$Kg/cm^2$
Description:	Pressure in the stack gas during baseline campaign
Source of data to be	AMS (Flow meter).
used:	
Value of data applied	Not applicable. We do not use this parameter to estimate expected emission
for the purpose of	reduction.
calculating expected	
emission reductions in	
section B.5	
Description of	Stack flow is measured by ANNUBAR device (multiple pressure differential
measurement methods	principle) with automatically compensates for actual stack pressure and
and procedures to be	temperature in order to normalize output data
applied:	
QA/QC procedures to	Regular calibrations according to vendor specifications and recognised
be applied:	industry standards. Staff will be trained in monitoring procedures and a reliable
	technical support infrastructure will be set up.
Any comment:	Measured during the complete lifetime of the project activity.
	Recorded every two seconds.

Data / Parameter:	Project N <sub>2</sub> O Concentration in the Stack Gas, NCSG <sub>project</sub>
Data unit:	mg N <sub>2</sub> O/ Nm <sup>3</sup> (converted from ppm if necessary)
Description:	N <sub>2</sub> O concentration in the stack gas for the project campaign
Source of data to be	AMS (Infrared gas analyzer) at Fosfertil Piaçaguera NAP 2.
used:	
Value of data applied	$505 \text{ mg N}_2\text{O}/\text{Nm}^3$
for the purpose of	
calculating expected	
emission reductions in	
section B.5	
Description of	N <sub>2</sub> O concentration is measured by on-line analyzer (Non Dispersive Infra Red
measurement methods	principle). A gas stream is continuously drawn from the stack by the sampling
and procedures to be	system under proper conditions (line is heat traced to avoid condensation), and
applied:	driven to the infrared cell. The device is set up to measure concentration and
	record the output electronically every 2 seconds.
QA/QC procedures to	Regular calibrations according to vendor specifications and recognised
be applied:	industry standards (EN 14181). Staff will be trained in monitoring procedures
	and a reliable technical support infrastructure will set up.
Any comment:	Measured during the complete lifetime of the project activity.
	Recorded every two seconds.





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Data / Parameter:	Project Operating Hours, OH <sub>project</sub>
Data unit:	Hours
Description:	Total operating hours for the project campaign
Source of data to be	Process control system at Fosfertil Piaçaguera NAP 2.
used:	
Value of data applied	4,512 hours
for the purpose of	
calculating expected	
emission reductions in	
section B.5	
Description of	Operated hours are measured by the consumption of ammonia registered. The
measurement methods	unit begins operation with the ammonia introduction in the NO converter and,
and procedures to be	the end of the operation is characterized with the cut of ammonia flow to this
applied:	reactor
QA/QC procedures to	Critical instruments are calibrated on a routinely basis according to the quality
be applied:	assurance system of the plant.
Any comment:	Measured daily during the complete lifetime of the project activity.

Data / Parameter:	Project Nitric Acid Production, NAP <sub>project</sub>
Data unit:	ton 100% HNO <sub>3</sub>
Description:	Total nitric acid production for the project campaign
Source of data to be	Production logs of Fosfertil Piaçaguera NAP 2.
used:	
Value of data applied	Nitric acid production is assumed to be constant, so that project emissions do
for the purpose of	not vary from year to year. The value of the nitric acid production used for the
calculating expected	calculation of expected emission reductions is: 95,400 ton HNO <sub>3</sub> /year
emission reductions in	
section B.5	
Description of	Daily production is measured by using a flow meter (Magnetic head principle),
measurement methods	and correcting by the average of several concentration checkups performed in
and procedures to be	analytical lab
applied:	
QA/QC procedures to	Mass balance at the end-product storage tanks is performed on a monthly basis;
be applied:	cross checking of production from mass balance with direct flow measurement
	is performed routinely. Critical instruments are calibrated on a routinely basis
	according to the quality assurance system of the plant
Any comment:	Measured daily during the complete lifetime of the project activity.





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Data / Parameter:	Project Emission Factor, EF <sub>n</sub>
Data unit:	ton N2O / ton 100% HNO3
Description:	Project emission factor calculated from monitored data for the project
	campaign
Source of data to be	Calculated from monitoring data.
used:	
Value of data applied	$0.00158 \text{ ton } N_2O/\text{ ton } HNO_3$
for the purpose of	
calculating expected	
emission reductions in	
section B.5	
Description of	Calculated from monitored data.
measurement methods	
and procedures to be	
applied:	
QA/QC procedures to	No QA/QC procedure is needed.
be applied:	
Any comment:	Project emission factor per unit of nitric acid produced will be calculated based
	on measurements of the nitric acid production, stack gas flow rate, N2O
	concentration, and the operating hours. All parameters will be measured during
	a complete campaign before project implementation to properly characterize
	baseline emissions factor.
	Calculated once at the end of the project campaign

Data / Parameter:	Project Campaign Length, CL <sub>n</sub>
Data unit:	Ton 100% HNO <sub>3</sub>
Description:	The project campaign length for the <i>nth</i> campaign $(CL_n)$ is defined as the nitric
	acid produced during the <i>nth</i> campaign (see project Nitric Acid Production)
Source of data to be	Production logs of Fosfertil Piaçaguera NAP 2.
used:	
Value of data applied	49,734 ton 100% HNO <sub>3</sub>
for the purpose of	
calculating expected	
emission reductions in	
section B.5	
Description of	Daily production is measured by using a flow meter (Magnetic head principle),
measurement methods	and correcting by the average of several concentration checkups performed in
and procedures to be	analytical lab.
applied:	
QA/QC procedures to	Mass balance at the end product storage tanks is performed on a monthly basis;
be applied:	cross checking of production from mass balance with direct flow measurement
	is performed routinely. Critical instruments are calibrated on a routinely basis
	according to the quality assurance system of the plant
Any comment:	Measured during a complete campaign before project implementation to



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properly characterize baseline emissions factor.
Calculated once at the end of the project campaign

Data / Parameter:	Project Gauze Supplier, GS <sub>n</sub>
Data unit:	Umicore
Description:	Gauze supplier for the project campaign
Source of data to be	Procurement offices of Fosfertil Piaçaguera NAP 2.
used:	
Value of data applied	Umicore
for the purpose of	
calculating expected	
emission reductions in	
section B.5	
Description of	Cover of supply contract or bill for gauzes for project campaigns, or equivalent
measurement methods	document to prove commercial transaction.
and procedures to be	
applied:	
QA/QC procedures to	None
be applied:	
Any comment:	Recording each campaign

Data / Parameter:	Project Gauze Composition, GC <sub>n</sub>
Data unit:	% precious metals (Platinum, Rhodium, Palladium)
Description:	Gauze composition for the project campaign
Source of data to be	Nitric plant procurement office and gauze Supplier technical service
used:	department.
Value of data applied	Pt 95%, Rh 5%
for the purpose of	
calculating expected	
emission reductions in	
section B.5	
Description of	Section of supply contract for gauzes that specifies the technical characteristics
measurement methods	agreed during baseline campaign. If necessary, additional data could be
and procedures to be	requested to supplier's technical service office in order to provide complete
applied:	technical profile of gauzes.
QA/QC procedures to	None
be applied:	
Any comment:	Recording each campaign



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Data / Parameter:	Emission Factor set by regulation, EF <sub>reg</sub>
Data unit:	kg N <sub>2</sub> O/ ton HNO <sub>3</sub>
Description:	Local and national regulations on N <sub>2</sub> O and NO <sub>X</sub> emissions
Source of data to be	Local and National Regulations
used:	
Value of data applied	Not applicable. We do not use this parameter to estimate expected emission
for the purpose of	reduction.
calculating expected	
emission reductions in	
section B.5	
Description of	At date of introducing or change of regulation.
measurement methods	
and procedures to be	
applied:	
QA/QC procedures to	No QA/QC procedure is needed.
be applied:	
Any comment:	None.

Data / Parameter:	Moving average emission factor, EF <sub>ma,n</sub>
Data unit:	kg N <sub>2</sub> O/ ton HNO <sub>3</sub>
Description:	Moving average of emission factor
Source of data to be	Calculated from campaign emissions factors
used:	
Value of data applied	Not applicable. We do not use this parameter to estimate expected emission
for the purpose of	reduction.
calculating expected	
emission reductions in	
section B.5	
Description of	Calculated as the average of the emission factors of each project campaigns.
measurement methods	
and procedures to be	
applied:	
QA/QC procedures to	No QA/QC is needed.
be applied:	
Any comment:	Calculated at the end of each project campaign

Data / Parameter:	Minimum Emission Factor , EF <sub>min</sub>
Data unit:	kg N <sub>2</sub> O/ ton HNO <sub>3</sub>
Description:	Minimum emission factor after ten campaigns
Source of data to be	Determined from campaign emission factors
used:	
Value of data applied	Not applicable. We do not use this parameter to estimate expected emission
for the purpose of	reduction.





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calculating expected	
emission reductions in	
section B.5	
Description of	Calculated from monitored data.
measurement methods	
and procedures to be	
applied:	
QA/QC procedures to	No QA/QC is needed
be applied:	
Any comment:	Calculated after end of ten campaigns

# **B.7.2** Description of the monitoring plan:

Fosfertil Piaçaguera NAP 2 is operated by a centralized automated control system, so staff is qualified and experienced at operating technical equipment to a high level of quality standards.

Furthermore, plant has been operating stack emissions units (f.e.  $NO_X$ ) for a number of years. The plant has access to specialized technical services available from the Fosfertil Technology Group.

All trainings required as consequence of the implementation of the CDM project activity were already developed and included as part of the ISO 9000 standard procedures.

The plant manager will be responsible for the ongoing operation and maintenance of the  $N_2O$  monitoring system. Operation, maintenance, calibration and service intervals will be according to the manufacturer specifications and incorporated into the management structure of ISO 9000 standard procedures.

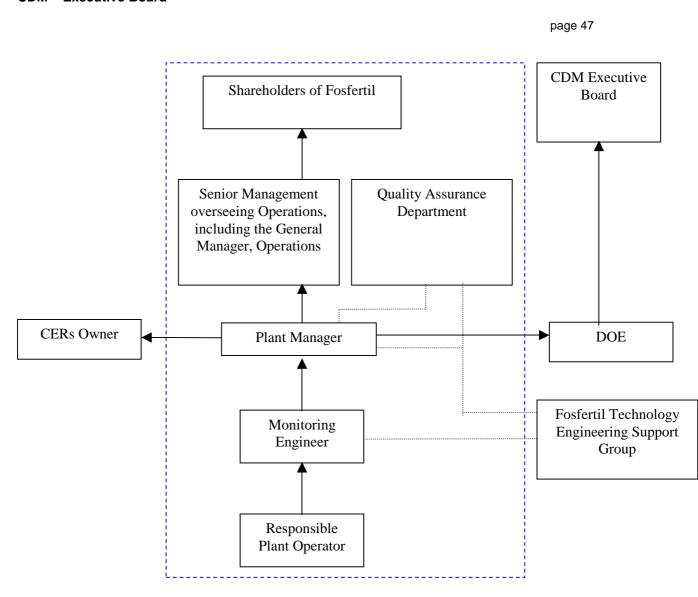
The proposed CDM project will be closely monitored, metered and recorded. The management and operation of the proposed nitrous oxide abatement project at the nitric acid plant will be the responsibility of the plant. The emission reductions will be verified by an independent entity, which will be a Designated Operational Entity (DOE). A regular (annual) reporting of the emission reductions generated by the project will be emitted to the CERs owner, coincidently with the DOE verification.

An illustrative scheme of the operational and management structure that will monitor the proposed CDM project activity is as follows:





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**Note:** the dashed line shows the operational and management structure boundaries of the proposed project.

The relation between the project operational and management structure, and other actors of the proposed CDM project activity, is described as follows:

- The responsible Plant Operator will be in charge of the supervision of the data acquisition system (DAS) that will be implemented to record plant operation data. Supported by the DAS, the Plant Operator will report the relevant data to the Monitoring Engineer.
- The Monitoring Engineer will be a member of the plant staff structure that will also be in charge of processing the data generated by the data acquisition system. The Monitoring Engineer will receive the relevant plant data from the responsible Plant Operator. These data will by entered into a spreadsheet especially designed for the monitoring plan.



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- The Plant Manager will be responsible to ensure that the CDM project activity at plant level is implemented in compliance with the PDD and other relevant standards. The Plant Manager will be assisted by the Quality Assurance Department which will conduct routine compliance audits. The Plant Manager will routinely report to the General Manager Operations as to the overall progress with the CDM project activity. At any time that the Plant Manager wants or needs to follow the implementation of the CDM project activity, he/she will ask for a report from the Monitoring Engineer. For every one year period, the Plant Manager will send a report which will basically be the monitoring plan spreadsheet to the CERs owner, as well as to the corresponding DOE.
- Fosfertil's Technology Engineering Support Group can at any time be used as a support function to the Monitoring Engineer in case of personnel loss or changes. The relevant Plant Manager and QA Department also have Fosfertil's Technology available as a resource for assistance when required.
- The DOE will then send the corresponding verification report to the CDM Executive Board in order to evaluate it and make able the issuance of the CERs.
- Shareholders of Fosfertil will receive annually from the plant manager, the same report sent to the DOE.

Considering the arguments and the schematic illustration above, a compliance with the monitoring methodology and the monitoring plan will be completely guaranteed.

# B.8 Date of completion of the application of the baseline study and monitoring methodology and the name of the responsible person(s)/entity(ies)

Date of completion of the application of this baseline and monitoring methodology to this project activity is: 12/02/2008 (prevision ).

The baseline and monitoring methodology has been applied by:

Walter Hügler, Nuria Zanzottera, and María Inés Hidalgo, MGM International Ltda. (not a project participant).

Tel: +54-11-5219-1230

e-mail: whugler@mgminter.com; nzanzottera@mgminter.com; ihidalgo@mgminter.com

# SECTION C. Duration of the project activity / crediting period

## C.1 Duration of the <u>project activity</u>:

# C.1.1. Starting date of the project activity:

13/02/2008, following completion of measurements to establish baseline emissions factor.

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C.1.2. Expected operational lifetime of the project activity:		
25 years-0 mor	nth.	
C.2 Choice	e of the <u>creditin</u>	ng period and related information:
C 2.1	D	
C.2.1.	<u>Kenewabie cr</u>	rediting period
	C.2.1.1.	Starting date of the first <u>crediting period</u> :
13/02/2008		
	C.2.1.2.	Length of the first <u>crediting period</u> :
7 years-0 mont	h.	
C.2.2.	Fixed crediting	ng period:
Not selected.		
	C.2.2.1.	Starting date:
N.A.		
	C.2.2.2.	Length:

N.A.



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# **SECTION D.** Environmental impacts

# **D.1.** Documentation on the analysis of the environmental impacts, including transboundary impacts:

Fosfertil Piaçaguera NAP 2 Nitrous Oxide Abatement Project involves the installation of secondary catalysts whose only purpose and effect is the decomposition of nitrous oxide once it is formed. After project implementation waste  $N_2O$  will be converted into  $N_2$  and  $O_2$  avoiding the high global warming effects of the GHG.

The installation of secondary catalysts has a positive environmental impact because it reduces  $N_2O$  emissions to the atmosphere and thereby results in cleaner overall air quality.

The project activity involves the installation of a secondary catalyst system inside the reactor immediately underneath the primary gauze system. The exhausted catalyst will be removed and replaced by the technology provider, who has developed the selected technology. No waste liquids, solids or gases are generated by using this technology. No further environmental impacts are expected.

Then, an Environmental Impact Assessment (EIA) is not necessary for this activity as it is stated in the national regulation. Fosfertil Piaçaguera NAP 2 fulfils the Operating License issued by the Environmental Agency of Sao Paulo State (Cetesb). NO<sub>X</sub> concentration in the stack gas of the nitric acid plant is lower than or equal at 200 ppm volume expressed as NO<sub>2</sub>.

D.2. If environmental impacts are considered significant by the project participants or the <u>host Party</u>, please provide conclusions and all references to support documentation of an environmental impact assessment undertaken in accordance with the procedures as required by the host Party:

No significant negative environmental impacts are expected from the implementation of the project activity. An environmental impact study is not required by Brazilian authorities

### **SECTION E. Stakeholders' comments**

# E.1. Brief description how comments by local stakeholders have been invited and compiled:

On September 11, 2003, the CIMGC, which is the Brazilian DNA issued the Resolution number 1, which was revised and complemented on October 18, 2005. This Resolution establishes some rules on how to conduct the local stakeholders consultation process. The stakeholders' process for the "Fosfertil Nitrous Oxide Abatement Project" was carried out according to these rules:

An invitation letter was sent to the main stakeholders, communicating Fosfertil's intention of submitting a greenhouse gas emission reduction project to the national and international authorities in order to generate carbon credits in the international market, called **Clean Development Fosfertil** –  $N_2O$ 



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**Abatement in Cubatão**. The letters were addressed to the main representatives of the below mentioned entities and the acknowledgements of receipt are available upon request and for validation purposes:

- Municipality of Cubatão;
- Alderman Chamber of Cubatão;
- Environmental City Secretariat of Cubatão;
- Environmental Secretariat of the State of São Paulo;
- Environmental Sanitation Tecnology Company (CETESB) Cubatão Region of the Environmental State Secretariat:
- Cubatão Justice Prosecution;
- Brazilian Forum of NGOs Forum Brasileiro de ONGs e Movimentos Sociais para o Meio Ambiente e Desenvolvimento;
- Federation of Industries of São Paulo Satate FIESP;
- Center of Industries of the São Paulo State Region of Cubatão;
- Workers Syndicate in Chemical, Pharmaceutical and Fertilizer Industries of Santos Lowlands (Sindquim).

The letter also informed that the full content of the Project Design Document, as well as the Annex 3 regarding its contribution to sustainable development, would be made available on the internet (at www.fosfertil.com.br, item "Responsabilidade Social/ Meio Ambiente/ Desenvolvimento Limpo") and asked recipients to provide their opinion, doubts or comments in writing about the project. The web page was kept open for more than 30 days after the reception of the letters by the invited entities.

# **E.2.** Summary of the comments received:

Comments from two Stakeholders were received. Stakeholders who gave their opinion are: CETESB and Environmental City Secretariat of Cubatão.

Both comments were positive and gave supported the project, remarking the project contribution to the environmental improvement.

CETESB sated that the Global Warming effect caused mainly by atmospheric emissions shows the need to take measures in order to mitigate and minimize the consequences of this effect. In that context CETESB understood the project implemented by Fosfertil as a pro active attitude that helps to improve the environment of the region; moreover no environmental license is required for this improvement. They finally, explained that the project must be submitted to analyse by Federal Ministry of Science and Technology

Environmental City Secretariat of Cubatão said that  $N_2O$  abatement project at Fosfertil site will contribute to improve the environmental quality of the region and will be possible due to the sale of CERs that helps to afford the investment costs. Environmental City Secretariat of Cubatão added that this is an important project of environmental improve in the sub-region of Cubatão.

## **E.3.** Report on how due account was taken of any comments received:

As the comments received were favourable, no adjustment in the project was necessary.



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# Annex 1

# CONTACT INFORMATION ON PARTICIPANTS IN THE PROJECT ACTIVITY

Organization:	ULTRAFERTIL S/A
Street/P.O.Box:	Estrada Eng. Plínio de Queiroz S/N
Building:	
City:	Cubatao
State/Region:	Sao Paulo
Postfix/ZIP:	11570-900
Country:	Brazil
Telephone:	
FAX:	
E-Mail:	
URL:	www.fosfertil.com.br
Represented by:	
Title:	
Salutation:	
Last Name:	Prado Santos
Middle Name:	
First Name:	Ricardo
Department:	
Mobile:	
Direct FAX:	(41) 643-3151
Direct tel:	(41) 641-1907
Personal E-Mail:	ricardosantos@fosfertil.com.br





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Organization:	Ecoinvest Carbon S.A
Street/P.O.Box:	Route de Florissant 13
Building:	
City:	Geneva-12
State/Region:	
Postfix/ZIP:	CH-1211
Country:	Switzerland
Telephone:	
FAX:	
E-Mail:	
URL:	
Represented by:	
Title:	
Salutation:	
Last Name:	Evans
Middle Name:	
First Name:	Alfred
Department:	
Mobile:	
Direct FAX:	41 22 5929100
Direct tel:	41 22 5929621
Personal E-Mail:	alfred.evans@bunge.com

# Annex 2

# INFORMATION REGARDING PUBLIC FUNDING

No public funds are available for the financing of the project activity. Therefore, Fosfertil will finance the project activity on the expectation of its approval.



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#### Annex 3

#### **BASELINE INFORMATION**

Baseline emissions will be calculated from an emission factor measured during a complete campaign before the implementation of the project activity, under normal operation conditions.

*Ex-ante* estimations of the key baseline parameters are listed in the following table:

Parameter	
Tail gas N <sub>2</sub> O concentration (ppm vol.)	1,286
Typical Nitric acid production output (ton 100% HNO <sub>3</sub> /year)	95,400
N <sub>2</sub> O baseline emission factor (kg N <sub>2</sub> O / ton 100% HNO <sub>3</sub> )	7.39
N <sub>2</sub> O destruction factor (%)	80
UNC (%)	5.79
Operating days	360

### Annex 4

## MONITORING INFORMATION

The current CDM project "Fosfertil Piaçaguera NAP 2 Nitrous Oxide Abatement Project" will measure on a quasi-continuous basis (uninterrupted sampling of flue gases with concentration and normalized flow analysis on short, discrete time periods) the  $N_2O$  mass flow leaving the Nitric acid plant through an Automated Measuring System (AMS<sup>8</sup>) using technologies and procedures in accordance with AM0034: "Catalytic reduction of  $N_2O$  inside the Ammonia burner of nitric acid plants".

Plant is currently ISO 9001/2000 and ISO 14001/2004 certified; certification documents will be available on site during validation for their review. The monitoring procedures (deployed as per the current monitoring plan and being an integral part of it), will be fully integrated into Piaçaguera NAP 2 Quality and Environmental Management System.

Fosfertil Piaçaguera NAP 2 is operated by a centralized automated control system, so staff is qualified and experienced at operating technical equipment to a high level of quality standards. Furthermore, Piaçaguera NAP 2 has been operating stack emissions units (f.e.  $NO_X$ ) since the installation of the plant. The plants have access to specialized technical services available from the Fosfertil Technology Engineering Support Group.

<sup>&</sup>lt;sup>8</sup> As per "Terms and definitions" of EN 14181:2004 (E), AMS definition is: Measuring system permanently installed on site for continuous monitoring of emissions. An AMS is a method which is traceable to a reference method. Apart from the analyzer, an AMS includes facilities for taking samples and for sample conditioning. This definition also includes testing and adjusting devices that are required for regular functional checks.



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All trainings required as consequence of the implementation of the CDM project activity were already developed and included as part of the ISO 9000 standard procedures.

The plant manager will be responsible for the ongoing operation and maintenance of the  $N_2O$  monitoring system. Operation, maintenance, calibration and service intervals will be according to the manufacturer specifications and international standards (see QA/QC section below), and incorporated into the management structure of ISO 9000 standard procedures.

The proposed CDM project will be closely monitored, metered and recorded. The management and operation of the proposed nitrous oxide abatement project at Fosfertil Piaçaguera NAP 2 will be the responsibility of the plant. The emission reductions will be verified at least annually by an independent entity, which will be a Designated Operational Entity (DOE). A regular (annual) reporting of the emission reductions generated by the project will be emitted to the CERs owner, coincidently with the DOE verification.

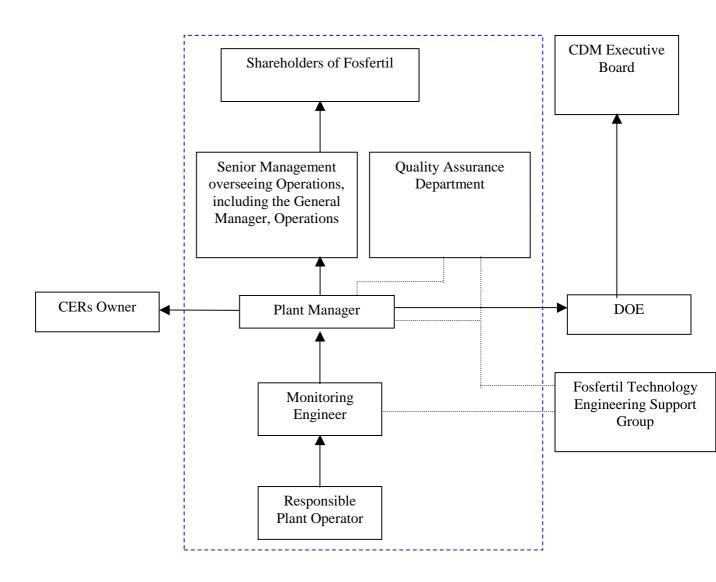
An illustrative scheme of the operational and management structure that will monitor the proposed CDM project activity is as follows:





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**Note:** the dashed line shows the operational and management structure boundaries of the proposed project.

The relation between the project operational and management structure, and other actors of the proposed CDM project activity, is described as follows:

• The responsible Plant Operator will be in charge of the supervision of automated measuring system (AMS) and the data acquisition system (DAS) that are installed to measure and acquire both emission and process data. Supported by the DAS, the Plant Operator will report the relevant data to the Monitoring Engineer.



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- The Monitoring Engineer will be a member of the plant staff structure that will also be in charge of processing the data generated by the data acquisition system. The Monitoring Engineer will receive the relevant plant data from the responsible Plant Operator. These data will by entered into a spreadsheet especially designed for the monitoring plan. The Project Developer will support the Monitoring Engineer at analyzing data and assure appropriate and consistent procedural application during report preparation.
- The Plant Manager will be responsible to ensure that the CDM project activity at plant level is implemented in compliance with the PDD and other relevant standards. The Plant Manager will routinely report to the General Manager Operations as to the overall progress of the CDM project activity. At any time that the Plant Manager wants or needs to follow the implementation of the CDM project activity, he/she will ask for a report from the Monitoring Engineer. For every one period, the Plant Manager will send a report which will basically be the monitoring plan spreadsheet to the CERs owner, as well as to the corresponding DOE.
- Fosfertil Technology Engineering Support Group can at any time be used as a support function to the Monitoring Engineer in case of personnel loss or changes. The relevant Plant Manager and QA Department also have Fosfertil Technology available as a resource for assistance when required.
- The DOE will then send the corresponding verification report to the CDM Executive Board in order to evaluate it and make able the issuance of the CERs.
- Shareholders of Fosfertil will receive annually from the plant manager, the same report sent to the DOE.

Tables on section B.7.1 of the PDD describe the parameters to be acquired and recorded as per the current monitoring plan, for both baseline campaign as well as (future) project campaigns. Furthermore, the baseline methodology requires that certain process parameters are monitored (to be compared vs the permitted operating conditions) during baseline campaign; such process parameters are also described in tables B.7.1. Only those  $N_2O$  measurements taken when the plant is operating within the permitted range will be considered during the calculation of baseline emissions.

All the relevant instrumentation to measure process parameters are calibrated on a routinely basis as per ISO system. The signals generated by these instruments are acquired and logged electronically by the Distributed Control System (DCS) of the plant. The specific data generated by the AMS is stored on the DCS every 2 seconds (after filtering for downtime and error readings). The DCS automatically provides an hourly average, which is then transferred onto a common spreadsheet (excel) for further analysis/calculations and reporting purposes. Actual emission reduction calculation will use values from such spreadsheet. Due to space constraints on the DCS hard-drive, from time to time, historical data will be archived on a separate hard drive or CDs, to be safeguard for at least 2 years. Raw (detailed) data will be accessible only through the DCS software platform, which insures the stored data cannot be manipulated.

All parameters measured during the baseline campaign will be archived in electronic and paper format during the entire crediting period and more two years after finishing the crediting period.



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All parameters measured during projects campaigns will be archived in electronic and paper format for at least two years.

# **Emission reduction calculations**

The amount of mass (tons) of  $N_2O$  that the project actually avoids from being vented to the atmosphere on each production campaign, expressed as Carbon Dioxide equivalent (or  $tCO_2e$ ), will be calculated by applying the following formulas:

$$BE_{BC} = VSG_{BC} \cdot NCSG_{BC} \cdot 10^{-9} \cdot OH_{BC}$$

Where:

 $BE_{BC}$  Total baseline emissions in the baseline measurement period, in, tN<sub>2</sub>O

 $VSG_{BC}$  Mean stack gas volume flow rate in the baseline measurement period, in Nm<sup>3</sup>/h

 $NCSG_{RC}$  Mean concentration of N<sub>2</sub>O in the stack gas in the baseline measurement period, in mg

 $N_2O/Nm^3$ 

 $OH_{BC}$  Number of operating hours in the baseline measurement period, in h

$$EF_{BL} = \frac{BE_{BC}}{NAP_{RC}} (1 - \frac{UNC}{100})$$

Where:

 $EF_{BL}$  Baseline emission factor, in  $tN_2O/tHNO_3$ 

 $NAP_{BC}$  Nitric acid production during the baseline campaign, in, tHNO<sub>3</sub>

UNC Overall measurement uncertainty of the monitoring system, in %, calculated as the

Combined uncertainty of the applied monitoring equipment

Project emissions are calculated from mean values of N<sub>2</sub>O concentration and total flow rate:

$$PE_n = VSG_n \cdot NCSG_n \cdot 10^{-9} \cdot OH_n$$

where:

 $PE_n$  Total Project emissions of the nth campaign, in tN<sub>2</sub>O

 $VSG_n$  Mean stack gas volume flow rate for the nth project campaign, in Nm<sup>3</sup>/h

 $NCSG_n$  Mean concentration of  $N_2O$  in the stack gas for the project campaign, in mg  $N_2O/Nm^3$ 

OH, Number of operating hours in the project campaign, in h

For the *nth* campaign, the campaign specific emission factor would be:

$$EF_n = \frac{PE_n}{NAP_n}$$





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where:

 $EF_n$  Emission factor calculated for the *nth* campaign, in kg  $N_2O/ton\ HNO_3$ 

 $PE_n$  Total Project emissions of the *nth* campaign, in tN<sub>2</sub>O

 $NAP_n$  Nitric acid production in the *nth* campaign, in ton 100% HNO<sub>3</sub>

Then,

$$ER_n = (EF_{BL} - EF_n) \cdot NAP_n \cdot GWP_{N_2O}$$

where

 $ER_n$  Emission reductions of the project for the *nth* campaign,  $tCO_2e$ 

 $EF_{BL}$  Baseline emission factor, in  $tN_2O/tHNO_3$ 

 $EF_p$  Project emission factor, applicable to the *nth* campaign, in  $tN_2O/tHNO_3$ 

 $NAP_n$  Nitric acid production during the *nth* campaign of the project activity, in, tHNO<sub>3</sub>

 $GWP_{N_2O}$  Global warming potential, of N<sub>2</sub>O set as 310 tCO<sub>2</sub>e/tN<sub>2</sub>O for the 1<sup>st</sup> commitment period

Following AM0034, several restrictions and adjustments will be applied to the formulas (above), among others:

1. All data series are filtered to eliminate mayericks and outliers.

The monitoring system will provide separate reading for  $N_2O$  concentration and gas flow for a define period of time (e.g. every hour of operation, i.e. an average of the measuring values of the past 60 minutes). Error readings (e.g. downtime or malfunction) and extreme values are eliminated from the output data series. Next, the same statistical evaluation that was applied to the baseline data series will be applied to the project data series:

- a) calculate the sample mean (x)
- b) calculate the sample standard deviation (s)
- c) calculate the 95% confidence interval (equal to 1.96 times the standard deviation)
- d) eliminate all data that lie outside the 95% confidence interval
- e) calculate the new sample mean from the remaining values
- 2. NAP (acid production) cannot exceed nameplate capacity of the plant

Nitric acid production will be compare to nameplate capacity. If nitric acid production at a given campaign is larger than nameplate, then emission reductions will be calculated ignoring data generated after production exceeds nameplate.

3. A moving average of the emission factors  $(EF_{ma})$  must be calculated

The campaign specific emissions factor  $(EF_n)$  for each campaign during the project's crediting period is compared to a moving average emission factor calculated as the average emission factor of the factors generated in the previous campaigns  $(EF_{ma,n})$ .



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To calculate the total emission reductions achieved in the *nth* campaign, the higher of the two values  $EF_{ma,n}$  and  $EF_n$  shall be applied as the emission factor relevant for that particular campaign  $(EF_p)$ .

4. A minimum project emission factor should also be determined  $(EF_{min})$ , defined as the lowest among the emission factor of the 10 first campaigns

After the first ten campaigns of the crediting period of the project, the lowest emission factor  $(EF_n)$  observed during those campaigns will be adopted as a minimum  $(EF_{min})$ . If any of the later project campaigns results in an  $EF_n$  that is lower than  $EF_{min}$ , the calculation of the emission reductions for that particular campaign shall used  $EF_{min}$  and not  $EF_n$ .

5. The emission factor to be applied for a particular campaign calculation  $(EF_p)$  must be the higher between the above mentioned moving average or the specific campaign emission factor (and not lower than minimum emission factor, after 10 campaigns)

This will be checked according to procedures detailed in steps 4 and 5 above.

6. The level of uncertainty (*UNC*) determined for the AMS installed at each plant, must be deducted from the baseline emissions factor.

The overall measurement uncertainty (*UNC*), calculated by summing in an appropriate manner (using gauss law of error propagation) all the relevant uncertainties arising from the individual performance characteristics of the AMS components, will be used to reduce the baseline emission factor, The following formulae will be applied:

$$EF_{BL} = EF_{BC} * (1 - \frac{UNC}{100})$$

7. If production at a given campaign is lower than normal ( $CL_{Normal}$ ), then the baseline is recalculated by ignoring the data generated after production exceeds normal campaign length.

The production at a given campaign will be compared to normal campaign length ( $CL_{Normal}$ ). If the length of each individual project campaign  $CL_n$  is shorter than the average historic campaign length, then  $EF_{BL}$  will be re-calculated by eliminating those  $N_2O$  values that were obtained during the production of tonnes of nitric acid beyond the  $CL_n$  (i.e. the last tonnes produced) from the calculation of  $EF_n$ .

Please note the specific calculations and adjustments to be followed according to the current monitoring plan are already described in detail in section B.6.1 "Explanation of methodological choices" of the Project Design Document.

# **Description of the AMS**



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Fosfertil Piaçaguera NAP 2 has installed continuous gas analyzers from the supplier ABB, model AO2000, while the specific module to measure N<sub>2</sub>O is a non-dispersive infrared called URAS 26. Below are the instruments descriptions as per manufacturer.

# Infrared Analyzer Module Uras26

#### Measurement Principle

Non-dispersive infrared absorption in the  $\lambda$  = 2.5–8  $\mu m$  wavelength range

Photometer to measure from 1 to 4 components with 1 or 2 beam paths and 1 or 2 receivers in each beam path

## Sample Components and Smallest Measurement Ranges

The Uras26 analyzer module has one physical measurement range per sample component. As an option, smaller measurement ranges can be electronically derived from the physical measurement range. The smallest range is measurement range 1.

The smallest measurement ranges shown in the following table are based on the first sample component in beam path 1.

Sample Compo- nent	Class 1 Range	Class 2 Range	Class 2 Range with Calibration Cell	Gas Group 1
CO	0- 50 ppm	0- 10 ppm	0- 50 ppm <sup>3</sup>	Α
CO:	0- 50 ppm	0- 5 ppm	0- 25 ppm <sup>23</sup>	A
NO	0- 75 ppm	0- 75 ppm	0- 75 ppm <sup>3</sup>	A
SO <sub>2</sub>	0- 100 ppm	0- 25 ppm	0- 25 ppm <sup>-3</sup>	A
N <sub>2</sub> O	0- 50 ppm	0- 20 ppm	0- 50 ppm <sup>th</sup>	A
CH.	0- 100 ppm	0- 50 ppm	0- 50 ppm <sup>2)</sup>	A
NH <sub>3</sub>	0- 500 ppm	0- 30 ppm	-	В
C <sub>2</sub> H <sub>2</sub>	0- 200 ppm	0- 100 ppm	0- 100 ppm	В
C <sub>2</sub> H <sub>4</sub>	0- 500 ppm	0- 300 ppm	0- 300 ppm	В
C <sub>2</sub> H <sub>0</sub>	0- 100 ppm	0- 50 ppm	0- 50 ppm <sup>th</sup>	В
C <sub>2</sub> H <sub>6</sub>	0~ 250 ppm	0- 100 ppm	0- 100 ppm <sup>2)</sup>	В
C <sub>2</sub> H <sub>6</sub>	0- 100 ppm	0- 50 ppm	0- 50 ppm <sup>2)</sup>	В
C <sub>2</sub> H <sub>10</sub>	0- 100 ppm	0- 50 ppm	0- 50 ppm <sup>2</sup>	В
C <sub>6</sub> H <sub>14</sub>	0- 500 ppm	0- 100 ppm	0- 100 ppm <sup>3</sup>	В
R 134a	0- 100 ppm	0- 50 ppm	0- 50 ppm <sup>2)</sup>	В
SF <sub>s</sub>	0-2000 ppm	0-1900 ppm	0-2000 ppm	В
H.O	0 1000 ppm	0- 500 ppm	0- 500 ppm	C

- 1) See price information
- Measurement range 1 the smallest is shown. The largest measurement range should be at least four times larger.

Other sample components on request.

The following data apply to measurement range 1 in a delivered analyzer module.

#### Sensitivity Drift

≤1% of measured value per week

#### Output Fluctuation (2 o)

- ≤ 0.2 % of span at electronic T90 time = 5 sec (Class 1) or
- = 15 sec (Class 2)

#### Detection Limit (4 a)

- $\leq$  0.4 % of span at electronic T90 time = 5 sec (Class 1) or = 15 sec (Class 2)
- Measurement Ranges

#### Quantity

1 to 4 ranges per sample component

#### Largest Measurement Range

0 to 100 Vol.-% or 0 Vol.-% to saturation or 0 Vol.-% to LEL Measurement ranges within ignition limits cannot be provided.

### Measurement Range Ratio

≤1:20

Measurement Ranges with Suppressed Zero-Point Electronic zero-point suppression or differential measurement based on a base level > 0 with flowing reference gas, max. suppression ratio of 1:10

### Measurement Range Switching

Manual; available external control or automatic

## Limit Value Monitoring

Limit values can be set during system configuration. The limit value signal (alarm) is output via the digital ports.

### Calibration

#### Zero-Point Calibration

With inert gas, e.g. N<sub>2</sub>, or with ambient air that is free of the sample component.

### End-Point Calibration

With gas-filled calibration cells (optional) or with test gas mixtures. It is recommended to verify the calibration cell set values once a year.

During calibration of a multi-component analyzer, possible cross-sensitivity and/or carrier gas corrections by internal or external measurement components are switched off. Therefore, corrected measurement components should be calibrated only using a test gas consisting of the measurement component and an inert gas like N<sub>a</sub>.

Influence Effects





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### Stability

#### Linearity Deviation

≤1% of span

Option: Linearization for automobile exhaust gas measurement according to EPA specifications

#### Repeatability

≤ 0.5 % of span

#### Zero Drift

< 1% of span per week;

for ranges smaller than Class 1 to Class 2:

≤3% of span per week

#### Influence Effects

#### Flow Effect

Flow rate in the 20-100 l/h range: within determination limits

### Associated Gas Effect/Cross Sensitivity

The knowledge of the sample gas composition is necessary for the analyzer configuration.

Selectivity measures to reduce associated gas effect (optional): Incorporation of interference filters, filter vessels or internal electronic cross-sensitivity correction or carrier gas correction for a sample component by other sample components measured with the Uras26.

# Infrared Analyzer Module Uras26

#### Temperature Effect

Ambient temperature in permissible range

- At zero-point: ≤1% of span per 10 °C; for ranges smaller than Class 1 to Class 2: ≤2 % of span per 10 °C
- On sensitivity with temperature compensation:
  - ≤ 3 % of measured value per 10 °C
- On sensitivity with thermostat effect at 55 °C (optional): ≤1% of measured value per 10 °C

#### Air Pressure Effect

- At zero-point: No effect.
- On sensitivity with pressure correction by means of integral pressure sensor: ≤ 0.2 % of measured value per 1 % barometric pressure change

The pressure sensor is located in the sample gas path if hoses are used as the internal gas lines.

If tubing is used for internal gas lines the pressure sensor is routed to the outside via a hose.

Pressure sensor working range: pats = 600-1250 hPa

### Power Supply Effect

24 VDC ± 5 %: ≤ 0.2 % of span

#### Dynamic Response

### Warm-Up Time

Approx. 30 minutes without thermostat; approx. 2 hours with thermostat

### 90% Response Time

 $T_{10} = 2.5$  sec for measurement cell length = 200 mm and sample gas flow = 60 l/h without signal damping (low pass filter). Low-pass time constant adjustable from 0 to 60 sec

#### Materials in Contact with the Sample Medium

#### **Gas Inlet Conditions**

#### Temperature

The sample gas dew point should be at least 5 °C below the ambient temperature throughout the sample gas path. Otherwise a sample gas cooler or condensate trap is required.

#### Inlet Pressure

p. = 2-500 hPa

Lower pressures require a sample gas pump and higher pressures require a pressure reducer.

#### Outlet Pressure

Atmospheric pressure

# Flow Rate

20-100 l/h

#### Corrosive Gases

Highly corrosive associated gas components, e.g. chlorine (Cl<sub>2</sub>) and hydrogen chloride (HCl), as well as gases or aerosols containing chlorine must be cooled or undergo prior absorption. Provide for housing purge.

#### Flammable Gases

The analyzer module is suitable for measuring flammable gases and vapors under atmospheric conditions ( $p_{aos} \le 1.1$  bar, oxygen content  $\le 21$  Vol.—96). Temperature Class: T4. The sample gas must not be explosive under normal conditions. If the sample gas is explosive in the event of a sample gas supply failure, then only seldom and briefly (in accordance with Zone 2). Pressure in the sample gas path in normal operation  $p_e \le 100$  hPa; in case of a sample gas supply failure the pressure must not exceed the maximum value  $p_e = 500$  hPa. The version with gas paths designed as stainless steel tubes should be selected and housing purge with  $N_2$  should be provided when measuring flammable gases and vapors. Before using the analyzer module the corrosion resistance against the specific sample gas must be checked.





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Analyzer (Sample Cells)
Tubing: Aluminum or gold-plated aluminum;
Window: CaF<sub>2</sub>, Option: BaF<sub>2</sub>;
Connectors: Rust- and acid-resistant steel 1.4571

Gas Lines and Connectors FPM hoses and PTFE tubing with stainless steel connectors; Option: Rust- and acid-resistant steel tubes 1.4571

### **Gas Connections**

Layout and Design
Gas ports on back (19-inch rack housing) or bottom (wall-mount housing) of the analyzer module with 1/8 NPT internal threads for commercially available adapters, e.g. Swagelok®.
See page 34 for connection drawing.

### **Electrical Connections**

System Bus 3-pin female plug External 24-VDC Power Supply 4-pin male plug Purge Gas

The purge gas should not contain any sample gas components.

### Power Supply

Input Voltage, Power Consumption 24 VDC ± 5 %, max. 95 W

#### Installation Site Requirements

Vibration

max.  $\pm 0.04$  mm at 5 to 55 Hz, 0.5 g at 55 to 150 Hz Slight transient effect on sample value in the region of the beam modulation frequency

Ambient Temperature

Operation: +5 to +40/45 °C when installed in housing with/without electronics module; Storage and transport: -25 to +65 °C

## Good monitoring practice and performance characteristics

Regarding QA/QC, the European Norm EN 14181:2004, which is recommended as guidance regarding the selection, installation and operation of the AMS under Monitoring Methodology AM0034, stipulates three levels of Quality Assurance Levels (QAL), and one Annual Surveillance Test (AST):

QAL1: Suitability of the AMS for the specific measuring task.

The suitability evaluation and its measuring procedure are described in ISO 14956:2002 "Air quality – Evaluation of the suitability of a measurement procedure by comparison with a required measuring uncertainty". Using this standard, it shall be proven that the total uncertainty of the results obtained from the AMS meets the specification for uncertainty stated in the applicable regulations (f.e. EU Directives 2000/76/EU or 2001/80/EU). Since European regulations do not yet cover the measurement of N<sub>2</sub>O at nitric acid plants, there is no official specification for uncertainty available. Then, considering official specification of uncertainties defined for equivalent pollutants (f.e. NOx, SO<sub>2</sub>) as per EU regulations, a 20% of the ELV (Emission Limit Value, in this case taken as the actual test concentration or calibration gas) has been considered by the equipment manufacturer as the required measurement quality for N<sub>2</sub>O, for purpose of expanded uncertainty calculations. The specific performance characteristics of the monitoring system chosen by the project shall be listed in the Project Design Document, as per AM0034. Then, tables below indicate such characteristics as per the corresponding QAL 1 report (extracted from the original QAL1 Report of each instrument)

The QAL1 Report for flow meter, Digimat Sonda 6; shows the following:





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Table 1 presents average values obtained for draining speeds in the duct and its respective uncertainties

Table 1: Average speeds in the transversal section

Surveyed	Average Speed	Specific Mass	Expanded Uncertainty
	(m/s)	$(kg/m^3)$	(%)
Horizontal	31.2	0.83	2.8
Vertical	30.3	0.83	2.6

Table 2 shows the average values obtained for flow in the duct and its respective uncertainties

Table 2-Average flow in the transversal section

Flow	Expanded Uncertainty
$(m^3/s)$	(%)
13.59	2.8

Table 3 shows the value of the discharge coefficient calculated and its uncertainty

Table 3-Discharge coefficient of the flow meter

Discharge coefficient K	Expanded uncertainty
(-)	(%)
0.656	2.9

The expanded uncertainties declared are based on its standard uncertainties combined, multiplied by the coverage factor K equal to2; considering 95% possibility of coverage approximately.

The QAL1 Report for URAS26 shows the following:







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Selectivity H2O	0,00	ppm
Selectivity others (largest sum)	1,86	ppm
Lack of fit	0,17	ppm
Drift	83,14	ppm
Pressure dependence	0,00	ppm
Temperature dependence	33,41	ppm
Flow dependence	0,00	ppm
Voltage dependence	0,00	ppm
Repeatability	0,21	ppm
Uncertainty of response factors	0,00	ppm
Response time	43	seconds
Origin of data	Test report	
Long-term drift of calibration cell	2,77	ppm
Origin of data	Article in UmweltMagazin, 2001	
Combined uncertainty of SRM	10,33	ppm
Standard Reference Method (SRM), Reference	Gas chromatography, VDI 2469	
ncertainty of cylinder gas	16,00	ppm
rigin of data	Datasheet of gas supplier	

The complete EN 14181: 2004 QAL1 reports are provided by the equipment manufacturers considering the performance characteristics as measured by a qualified Technical Inspection Authority (such as the German TÜV) and the specific installation characteristics and site conditions at the plant. The QAL1 reports confirm the  $N_2O$  analyzers (AO 2000 Series with analyzer module Uras 14/26) are suitable to perform the indicated analysis ( $N_2O$  concentration), and provide a conservative estimation (meaning actual performance would be better) for expanded uncertainty. The complete QAL1 reports are available for validation.

The overall measurement uncertainty (*UNC*) is calculated by summing in an appropriate manner (using gauss law of error propagation) all the relevant uncertainties arising from the individual performance characteristics of the AMS components (then  $UNC = ((N_2O \text{ Analyzer uncertainty})^2 + (Flow meter uncertainty)^2)^{1/2}$ ). The overall measurement uncertainty will be available for the validation of the project activity.

### QAL2: Validation of the AMS following its Installation.

The next level of quality assurance prescribed on EN14181:2004 (QAL2), describes a procedure for the determination of the calibration function and its variability, by means of certain number of parallel measurements (meaning simultaneously with the AMS), performed with a Standard Reference Method (which should be a proven and accurate analytical protocol as per relevant norms or legislation). The variability of the measured values obtained with the AMS is then compared with the uncertainty given by the applicable legislation, if the measured variability is lower than the permitted uncertainty, it is concluded the AMS has passed the variability test. Since (as explained above), official uncertainty is not

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 $<sup>^{9}</sup>$  Considering EN 14181 does not specify what SRM to use for each specific compounds, there is controversy as to which method is suitable as SRM for  $N_2O$ , since the best available technology (and hence the most accurate instrument) is the actual online instrument which is the subject of calibration by this method.





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available, an appropriate level is determined based on those that do exist for similar pollutants and techniques (in this case 20% of ELV). As per international standards, there are two potentially suitable Standard Reference Methods: 1) bench scale gas chromatography as per VDI standard 2469 or 2) Nondispersive Infrared Method, as per ISO 21258 (draft).

The testing laboratories performing the measurements with the Standard Reference Method shall have an accredited quality assurance system according to EN ISO/IEC 17025 or relevant (national) standards. Fosfertil is in the process of selecting a suitable testing laboratory to conduct the QAL2 tests which are planned to take place in late September or early October, 2007 during the baseline measurements. The QAL2 report from the certified lab will be available for DOE review during the verification. Any data collected previous to the reception of the QAL2 lab report will be corrected through proper application of the calibration function.

As condition precedent for a QAL2 test, it is required that the AMS has been correctly installed and commissioned, considering (for example) that the AMS is readily accessible for regular maintenance and other necessary activities and that the working platform to access the AMS allows for parallel sampling. The AMS unit at Fosfertil Piaçaguera NAP 2 was installed by qualified contractors under the direct supervision of the equipment manufacturers, considering both relevant Brazilian and international standards. The Plant Manager, as well as members of Fosfertil Technology Engineering Support Group, actively supervised all phases of installation, from system design to commissioning.

## QAL3: Ongoing quality assurance during operation.

Procedures described at QAL3 of EN 141181: 2004 check for drift and precision, in order to demonstrate that the AMS is in control during its operations so that it continues to function within the required specification for uncertainty. This is achieved by conducting periodic zero and span checks on the AMS, and evaluating results obtained using control charts. Zero and span adjustments or maintenance of the AMS may be implemented, as result of such evaluation. The implementation and performance of the QAL3 procedures given in this standard are the responsibility of the plant (or AMS) owner.

The standard deviation according to QAL3 will be calculated by the equipment manufacturer based on equipment performance characteristics and field conditions for Fosfertil Piacaguera NAP 2. Calculation spreadsheets from the suppliers will be available for validation. The data is used to monitor that the difference between measured values and true values of zero and span reference materials are equal or smaller than the combined drift and precision value of the AMS multiplied by a coverage factor of 2 (2 times standard deviation of AMS, as described in QAL3 section of EN14181) on a weekly basis, with the aid of Shewart charts. Documented calibration procedure for weekly zero and span checks as well as resulting Shewart charts will be available on site for validation and future verifications.

All monitoring equipment will be serviced and maintained according to the manufacturer's instructions and international standards by qualified personnel (both Fosfertil Piaçaguera NAP 2 resources and the third parties involved during such activities). Maintenance and service logs will be well kept at Fosfertil Piaçaguera NAP 2 and available for auditing purposes.

AST: Annual Surveillance Test (ongoing quality assurance).





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The AST is a procedure to evaluate whether the measured values obtained from the AMS still meet the required uncertainty criteria, as evaluated during the QAL2 test. As the QAL2, it also requires a limited number of parallel measurements using an appropriate Standard Reference Method. An AST should be performed to the plant's AMS at least once every 5 years; its frequency depends on the relation between the total expected uncertainty of the AMS and the selected required uncertainty.